

Use of neutrons in biology and medicine

Jayne Lawrence

Pharmaceutical Science Division

King's College London

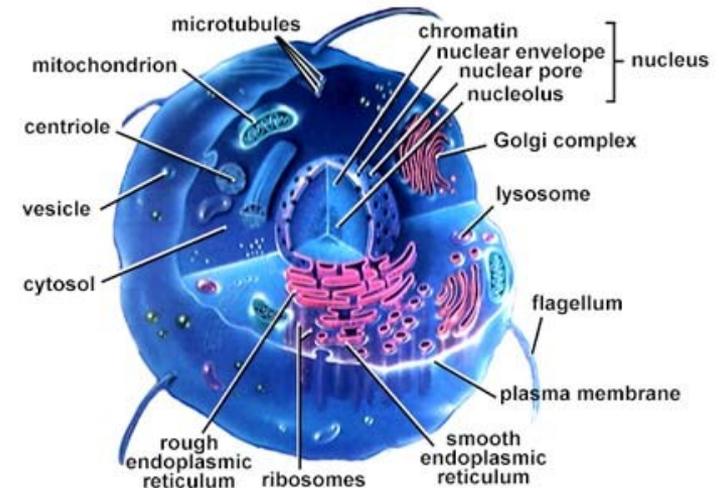
London

Useful reading

- Chapter 23 Neutron crystallography of proteins and Chapter 24 Molecular Biology in Methods of Experimental Physics Volume 23 Part C Neutron Scattering (1987) Academic Press Inc (London)
- Neutron scattering in biology – techniques and applications (2006) Springer-Verlag Berlin Heidelberg, (New York)
- Neutrons for biologists: a beginners guide, or why you should consider using neutrons JH Lakey J. R. Soc. Interface (2009) 6 S567-S573.
- <http://www.strubi.ox.ac.uk/people/gilbert/neutrons.html#spins>

Biology - study of living organisms

- structure, function, growth, origin, evolution, distribution and classification of living things
- neutrons particularly useful in the study of *structure*
- but why?



Neutrons a good probe for biological studies because.....

- Neutrons penetrate deeply into biological materials
- Neutrons do not damage biological materials
- Neutron wavelength \sim atomic dimensions
- Neutron energies \sim atomic motions
- Large dynamic range in space, time and energy
- Neutrons can distinguish between hydrogen and deuterium (basis of ***contrast variation***)
- Neutrons can detect the location of protons (essential in many biological processes)

Neutrons provide structural and dynamic data.....

- Molecular size and shape
- Membrane structure
- Atomic structure
- Molecular dynamics

However the technique suffers from.....

- Low resolution
- Low flux
- Longer acquisition times
- Requirement for relatively large samples

Scattering lengths

Element	Neutrons	X Rays
	$b \times 10^{13}$ (cm)	$b \times 10^{13}$ (cm)
H	-3.74	3.8
D	6.67	2.8
C	6.65	16.9
N	9.40	19.7
O	5.80	22.5
P	5.10	42.3
S	2.85	45.0
Mn	-3.60	70.0
Fe	9.51	73.0
Pt	9.50	220.0

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X-ray and neutron scattering

X-ray

Scattering proportional to Z

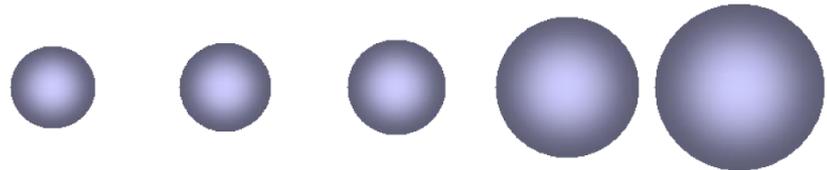
H B C O
1 3 4 8



Neutron

Scattering not proportional to Z

Al Si P Ti Fe
13 14 15 22 26



-3.74

5.30

6.65

5.80

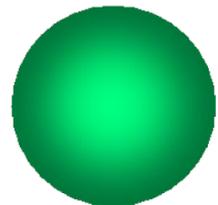
3.45

4.15

5.13

-3.44

9.45



X-ray and neutron scattering

X-ray

Scattering proportional to Z

Neutron

Scattering not proportional to Z

H

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C

O

Al

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P

Ti

Fe

1

3

4

8

13

14

15

22

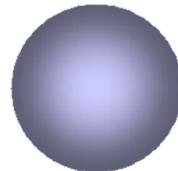
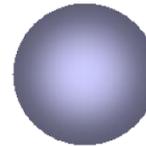
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6.67

5.30

6.65

5.80

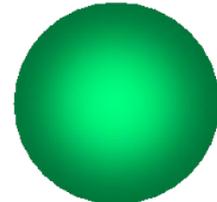
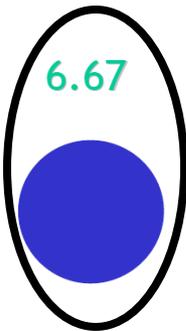
3.45

4.15

5.13

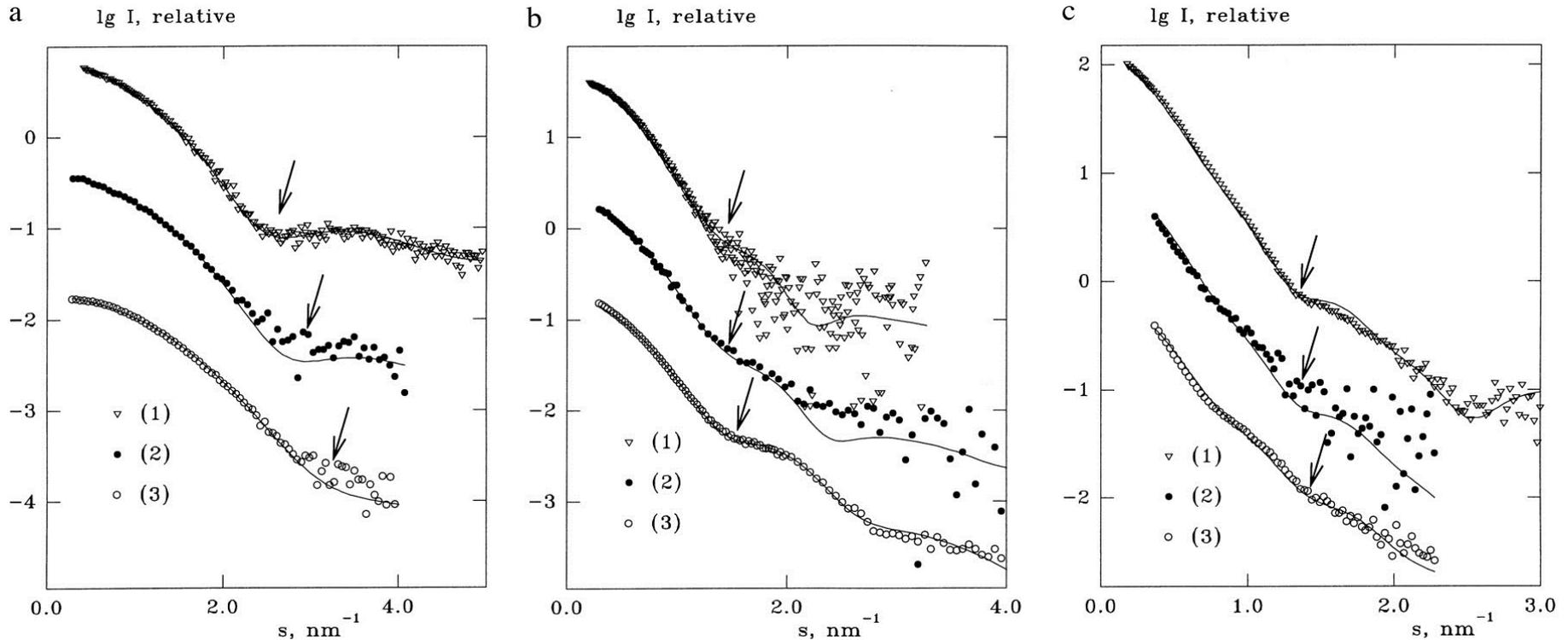
-3.44

9.45



Contrast variation

- In the absence of any specific isotope labelling, large range of contrasts possible through the use of mixture of D_2O/H_2O

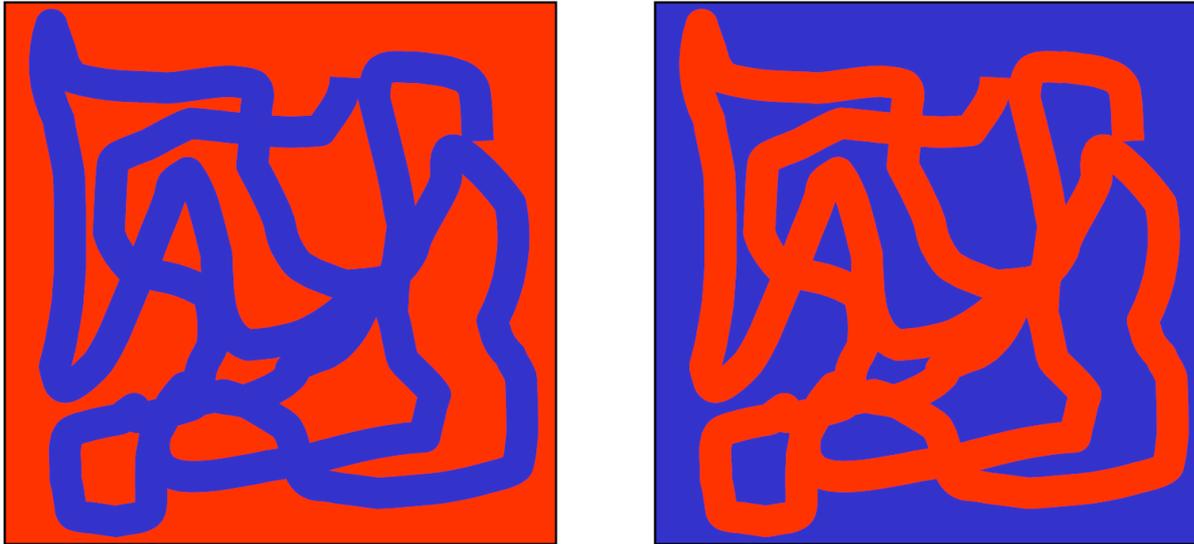


SANS profiles for (a) lysozyme (b) thioredoxine reductase and (c) ribonucleotide reductase. Sets of data (1-3) correspond to sample measured by x-rays, neutrons (sample dispersed in H_2O) and neutrons, (sample dispersed in D_2O).

Contrast variation

- Remember not all protons exchangeable – may take time for exchange to occur – scattering length density changes with solvent D_2O/H_2O composition.
- To distinguish between the components of a system in which the scattering length densities are all similar require selective labelling of specific components - normally with deuterium
- Contrast variation was first exploited in the classic experiments on the 30S sub-unit of the ribosome (M. S. Capel et al *Science* (1987) 238, 1403-1406) and σ -factor and core enzyme of *E. coli* RNA polymerase (Lederer et al *J. Mol. Biol.* (1991) 219, 747-755) where labelling was exploited to determine by triangulation the distances between the protein components.
- It is essential to ensure that replacement of hydrogen with a deuterium does not alter the properties of the system under study

Contrast variation



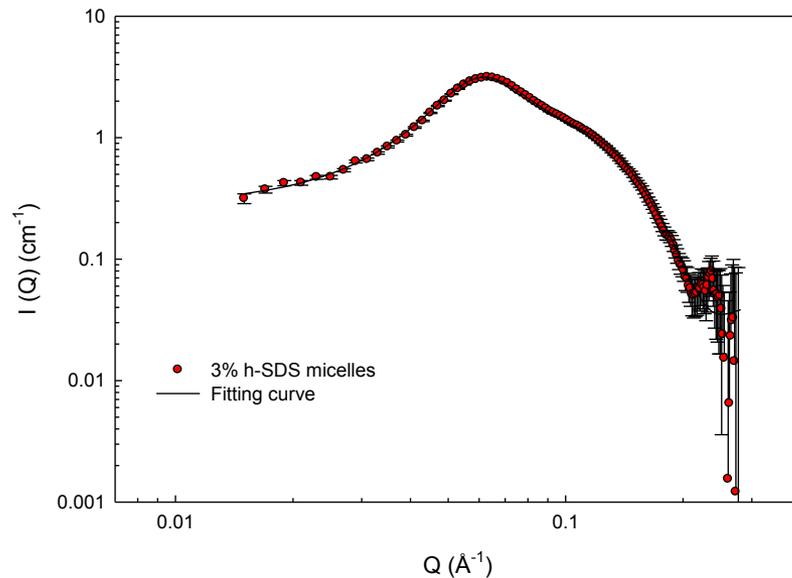
Two structures give the same scattering, although the incoherent scattering may be different

$$\frac{d\Sigma}{d\Omega}(\vec{q}) \propto (\rho_1 - \rho_2)^2$$

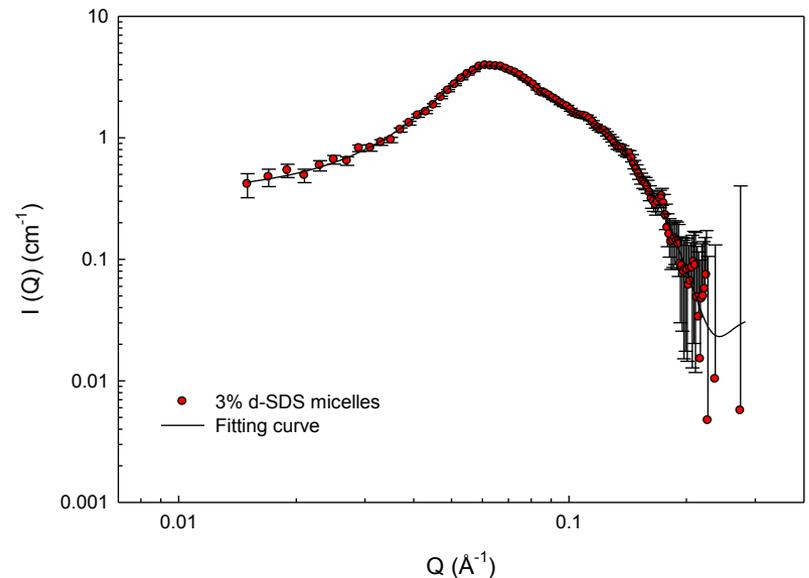
Substitution of deuterium for hydrogen

SANS profiles and corresponding fits obtained using ellipsoidal model and Hayter Penfold structure factor 3% w/v solutions of

(a) h-SDS in D₂O

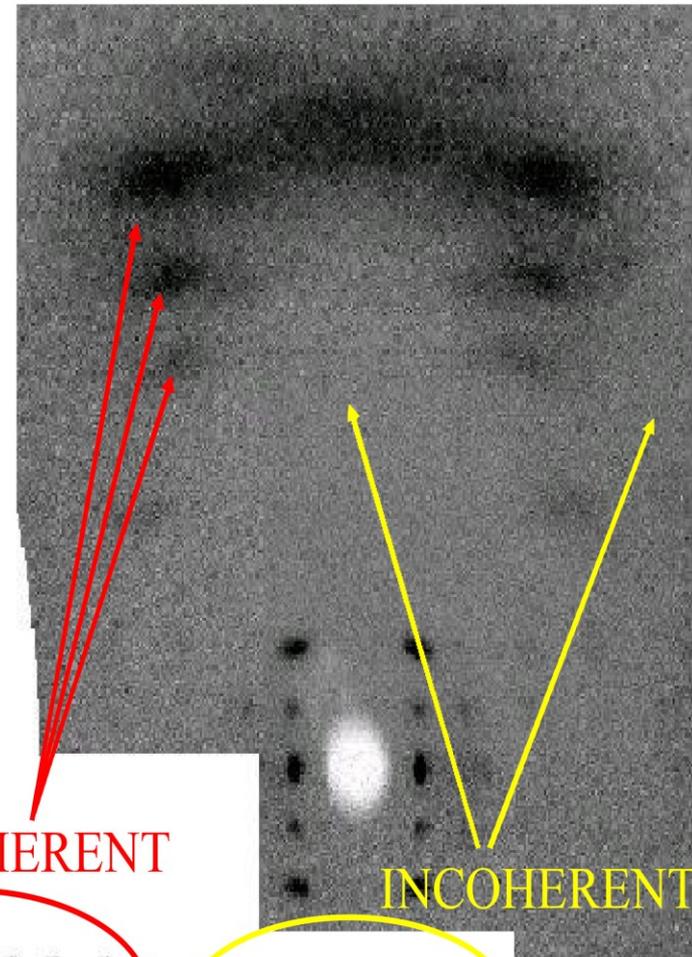
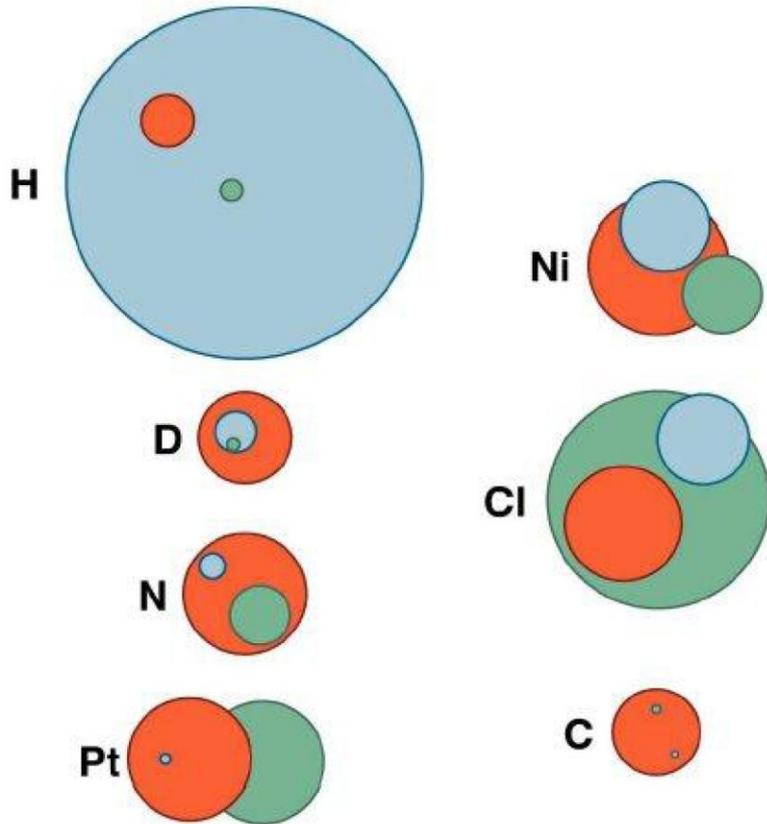


(b) d-SDS in H₂O



	R_{core} (Å)	axial ratio of core	shell thickness (Å)	minor radius (Å)	major radius (Å)	axial ratio	Charge/Z	INV Debye/k
h-SDS	16.5	1.49	3.3	19.8	27.8	1.40	23.0	0.041
d-SDS	16.5	1.45	3.2	19.8	27.3	1.38	23.5	0.041

Incoherent scattering



$$\frac{d\sigma}{d\Omega} = \langle b \rangle^2 \sum_{i,j} e^{-i\vec{Q} \cdot (\vec{R}_i - \vec{R}_j)} + (\langle b^2 \rangle - \langle b \rangle^2) N$$

Structure determination using neutron scattering

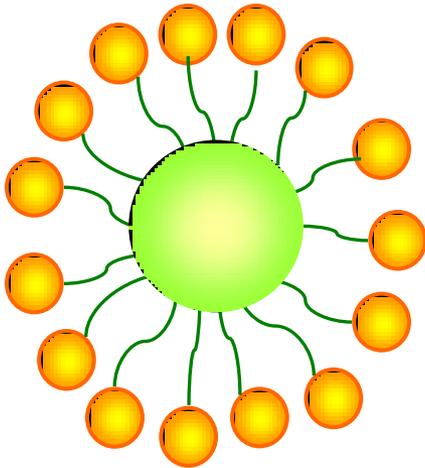
- Small angle neutron scattering
- Low resolution crystallography
- High resolution crystallography
- (Fibre diffraction)

- Membrane diffraction
- Reflectometry

Small angle neutron scattering

- Extensively used to characterize nanostructures and hierarchical structures of materials ranging from 1 to 400 nm
- Yields low resolution information on shape
- Contrast variation and labelling can be used to provide important information on structure in multi-component systems
- Numerous examples of SANS being used for biological and medical questions including:-
 - protein-surfactant interactions,
 - light-induced structural changes in pea thylakoids,
 - the solution structure of human proliferating cell nuclear antigen
 - biomineralization
 - gene delivery vehicles.....

Microemulsions as drug delivery vehicles

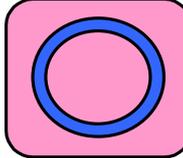
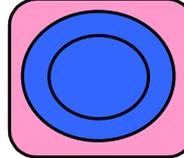


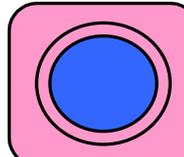
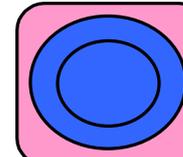
Hypothesis – oil core acts as an additional locus of solubilisation of drug in the aggregate

Observation – small molecular volume oils exhibiting a good capacity for drug do not results in microemulsions with increased capacity for drug

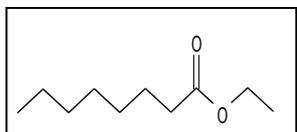
Understanding – provided by SANS

Microemulsions as drug delivery vehicles

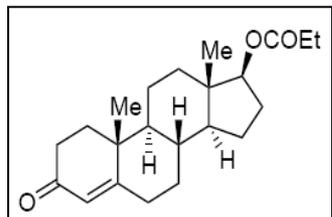
Compositions	Core	Shell	Droplet
■ Hydrogenated ■ deuterated			
Oil	h-Oil	h-Oil	h-Oil
Surfactant (SAA)	d-SAA	d-SAA	h-SAA
Solvent	D ₂ O	H ₂ O	D ₂ O

Compositions	Core	Shell	Droplet
■ Hydrogenated ■ Deuterated			
Oil	d-Oil	d-Oil	d-Oil
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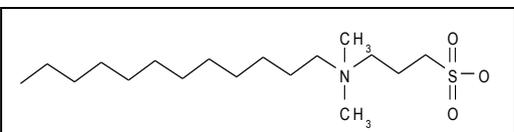
SANS data and fits of oil-in-water microemulsions with & without drug



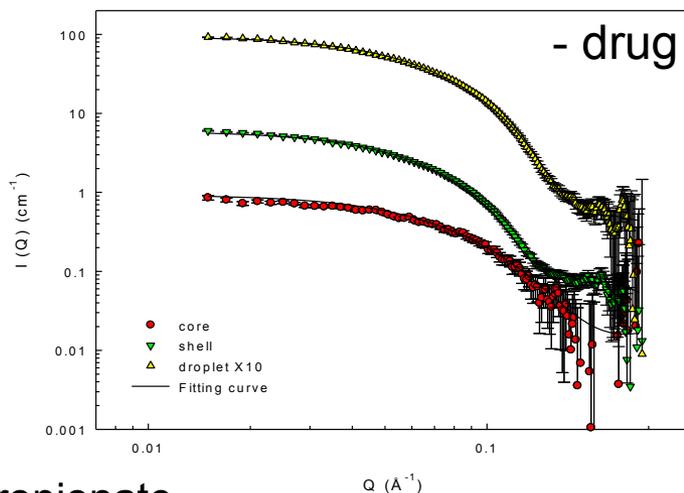
oil = ethyl caprylate



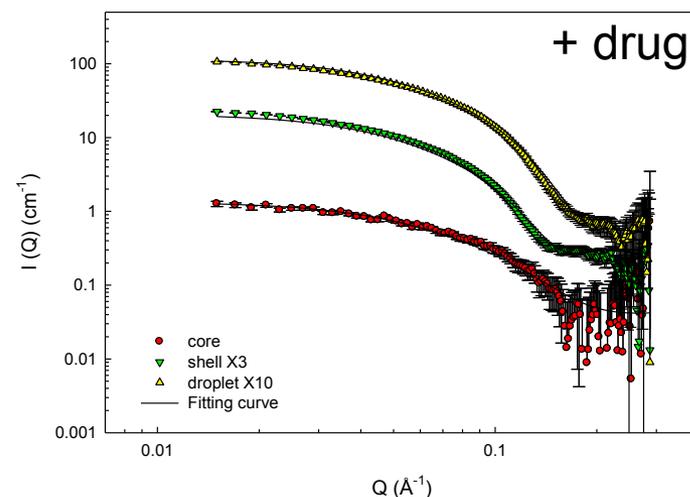
drug = testosterone propionate



surfactant = *N,N*-dimethyldodecylammoniumpropanesulfonate (DDAPS)



Data fitted using core-shell ellipsoid and hard sphere structure factor



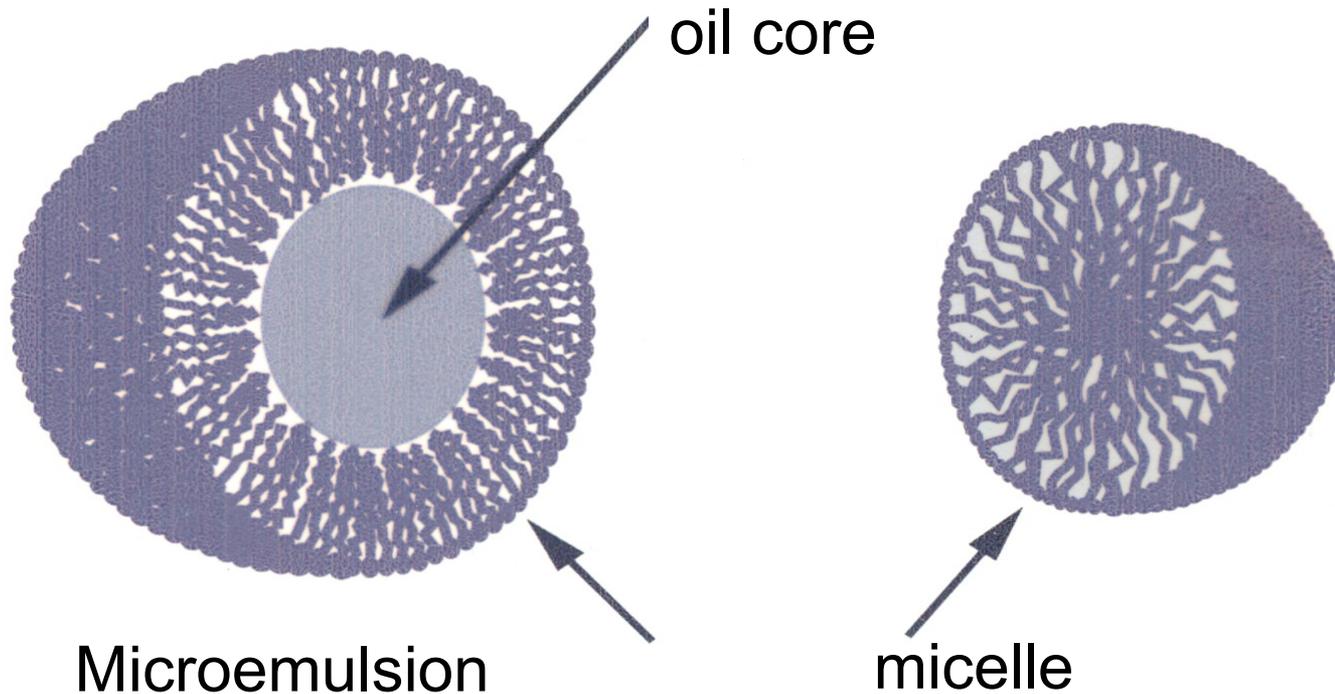
SUMMARY OF RESULTS

$$I(Q) = n \times P(Q) \times S(Q)$$

Samples	Shell thickness (Å)	Minor radius (Å)	Major radius (Å)	Axial ratio	P_{oil}	P_{drug} in shell
Microemulsion no drug	18.9	25.3	59.2	2.3	0.57	-
Microemulsion with drug	17.2	23.6	58.5	2.5	0.53	0.24

Effect of oil on drug solubilisation

Larger molecular volume oils formed a distinct core in the centre of the microemulsion droplet, whereas smaller oils mixed intimately with the surfactant tails

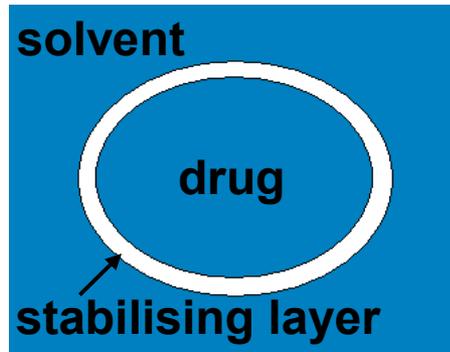


oil = extra locus of drug solubilisation?

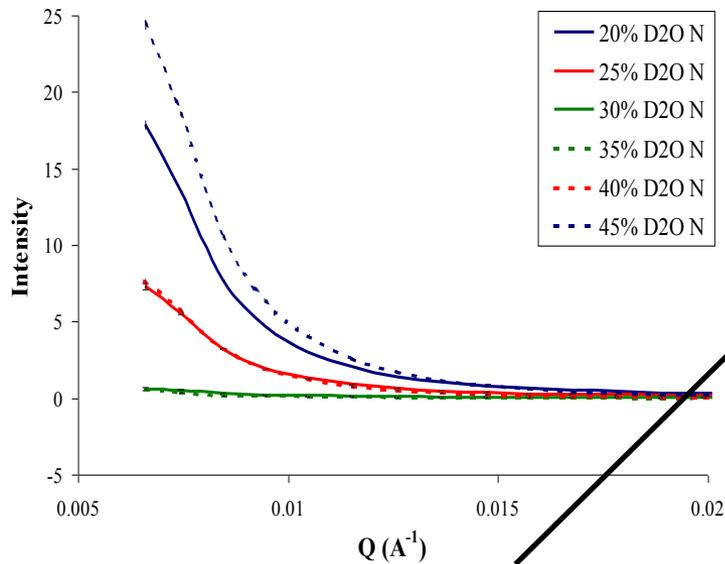
Nanosuspomicroemulsions

- Contain both drug nanoparticles (~ 300 nm) and microemulsions (~ 18 nm): able to deliver two different drugs in one formulation
- '*Stable*' version of a suspoemulsion (currently used in agrichemical industry)
- Prepared from simple mixing of nanoparticles & microemulsions
- *In situ* measurement of microemulsion stability was possible using neutrons, preparation cloudy in appearance

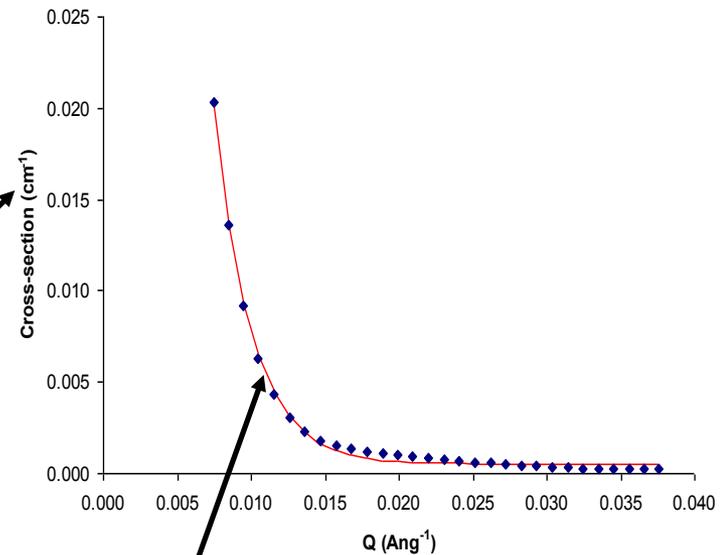
Contrast matching of drug nanoparticles



- nanoparticles dispersed in H₂O/D₂O mix that makes the nanoparticles 'invisible' to neutrons
- (very weak) scattering seen only from stabilising layer



extremely low scattering

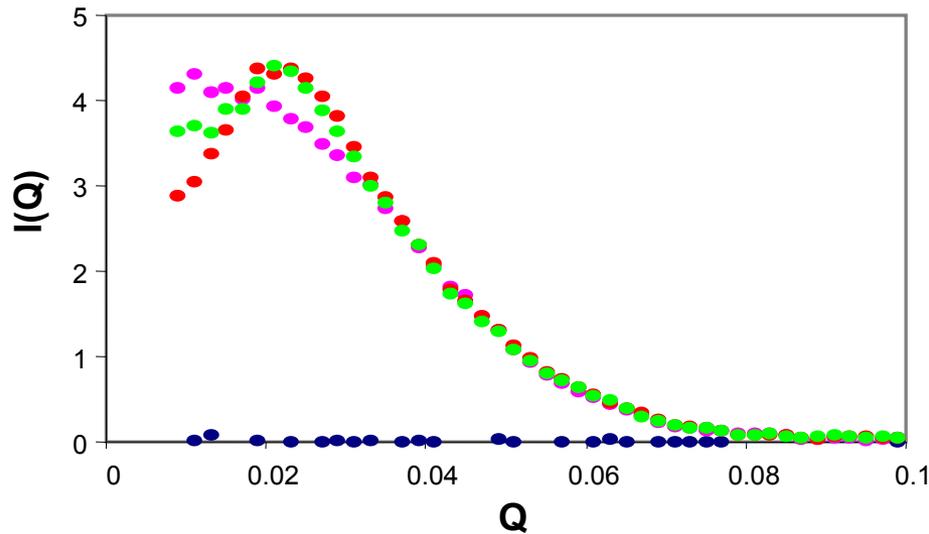


very low Q

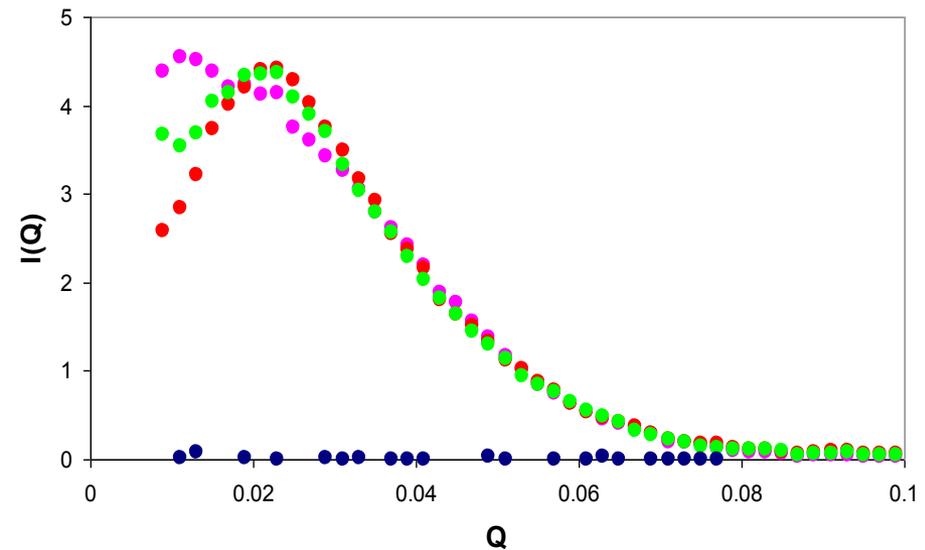
Nanosuspomicroemulsions

2.4 vol% of Brij 97 m/e & 2.4 vol% of SDS stabilised griseofulvin nanoparticles

no testosterone propionate



testosterone propionate

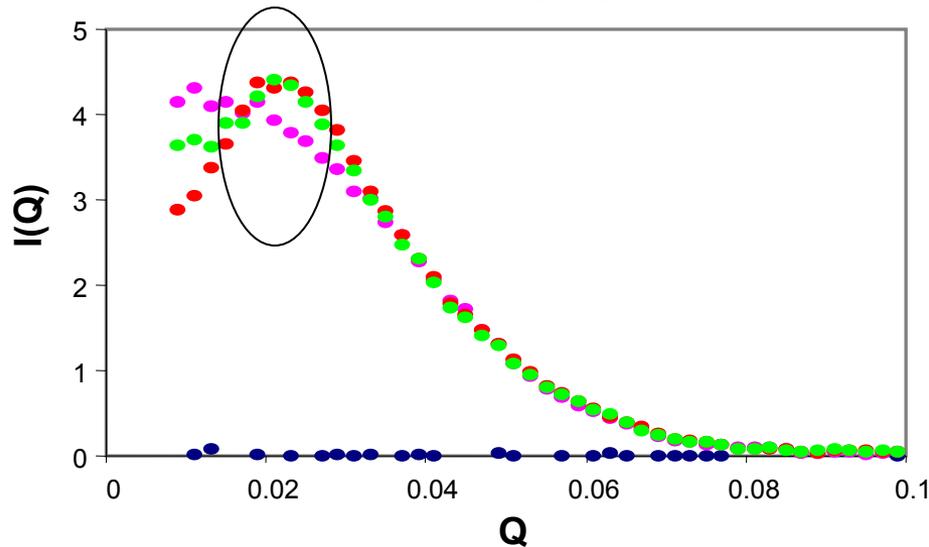


- nanoparticles
- microemulsions before contact with nanoparticles
- microemulsions immediately after contact with nanoparticles
- microemulsions after 24 h contact with nanoparticles

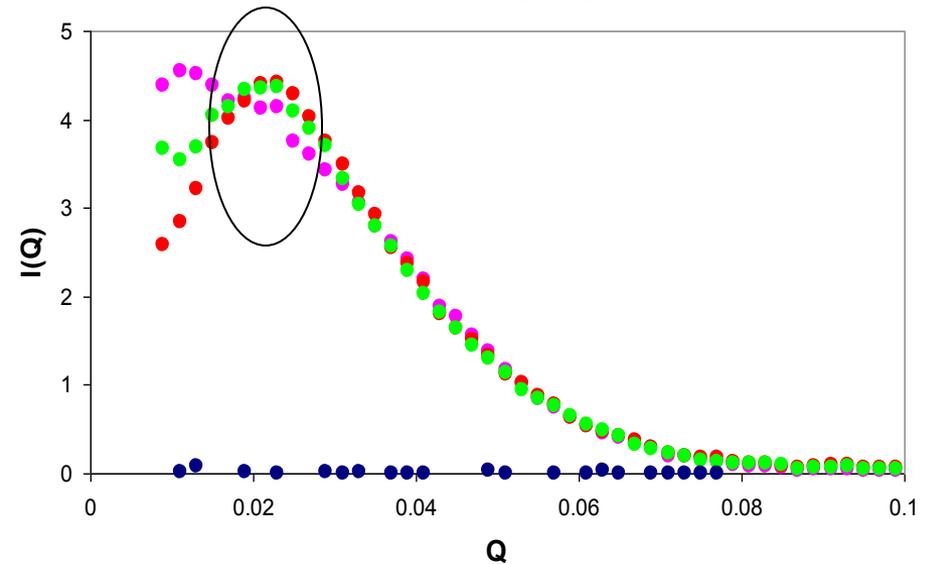
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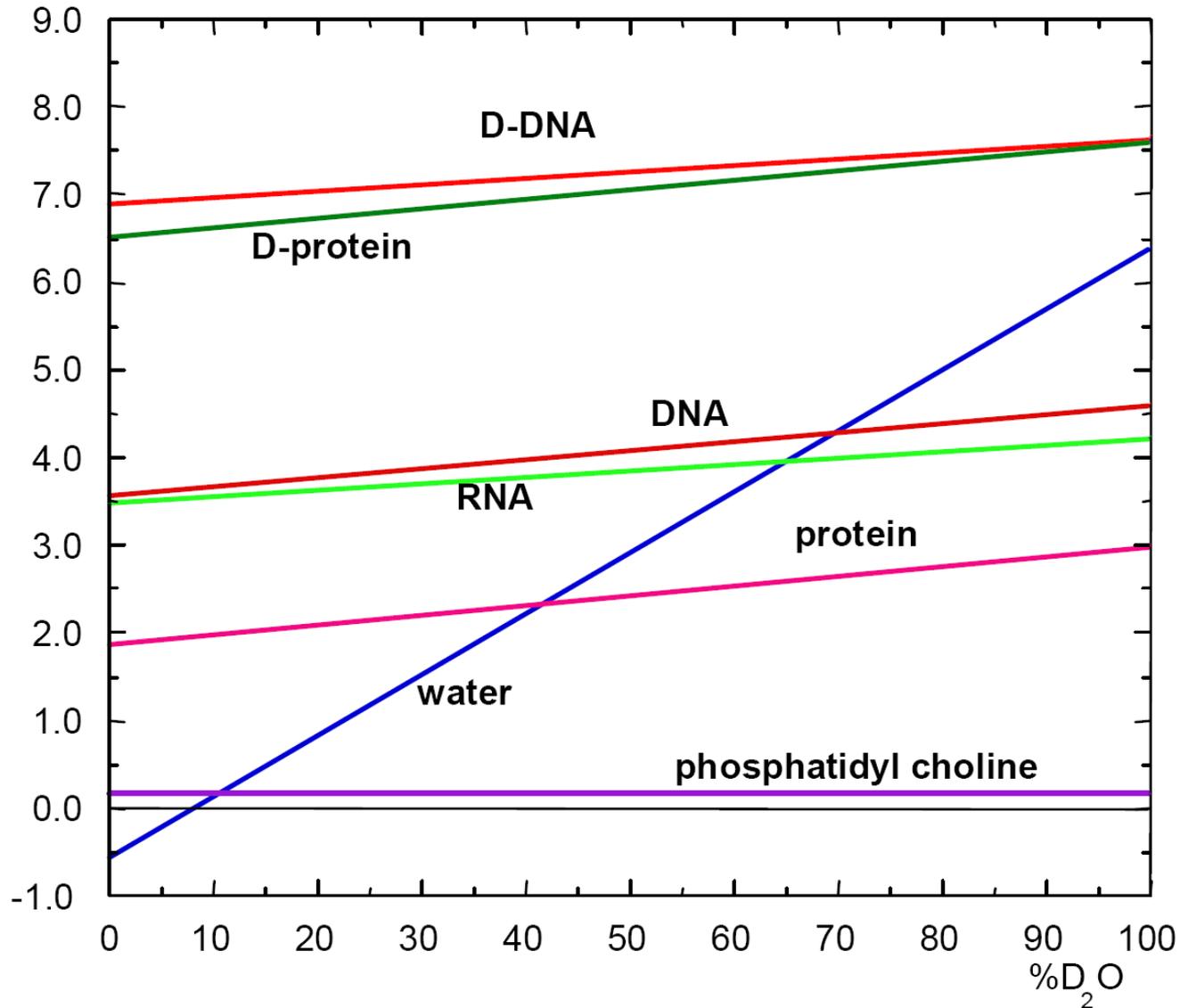
testosterone propionate



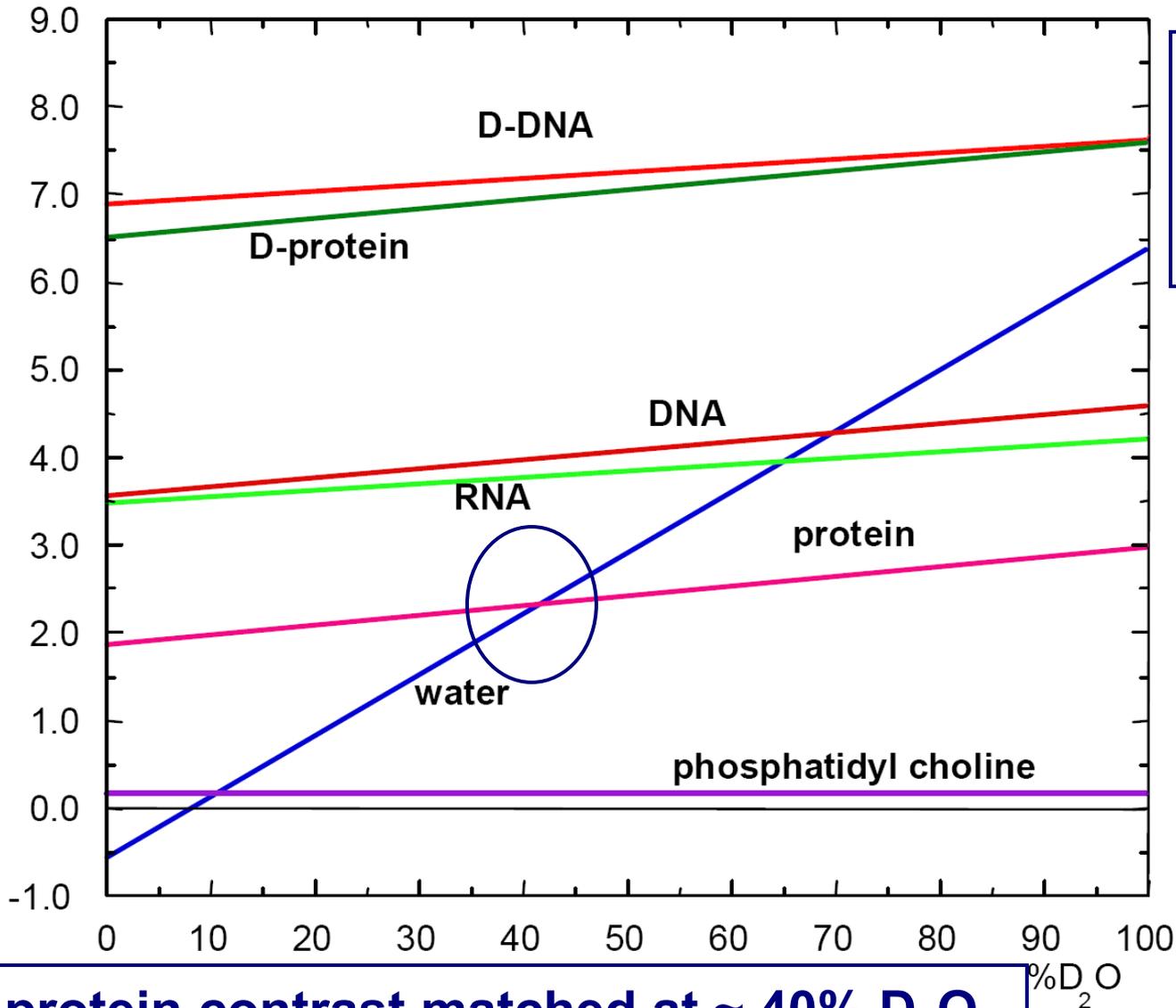
- nanoparticles
- microemulsions before contact with nanoparticles
- microemulsions immediately after contact with nanoparticles
- microemulsions after 24 h contact with nanoparticles

$S(Q)$, interaction peak due to exchange of charged SDS from nanoparticles to m/e

Variation in scattering length density with percent D₂O



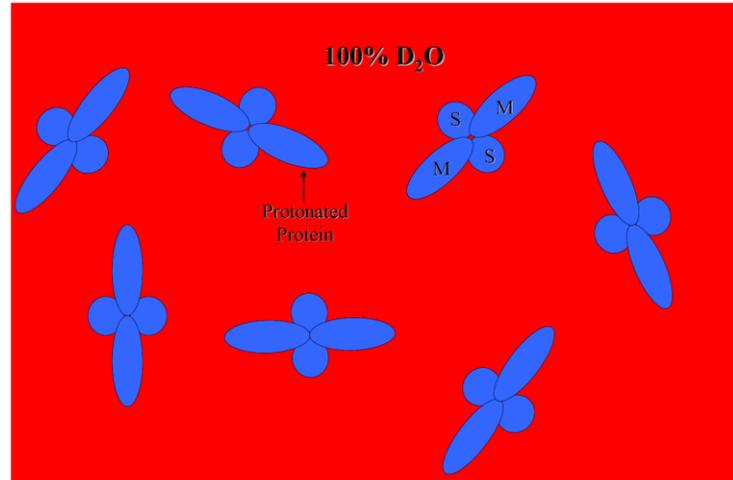
Variation in scattering length density with percent D₂O



NB Proteins studied at 80% rather than 100% deuteration

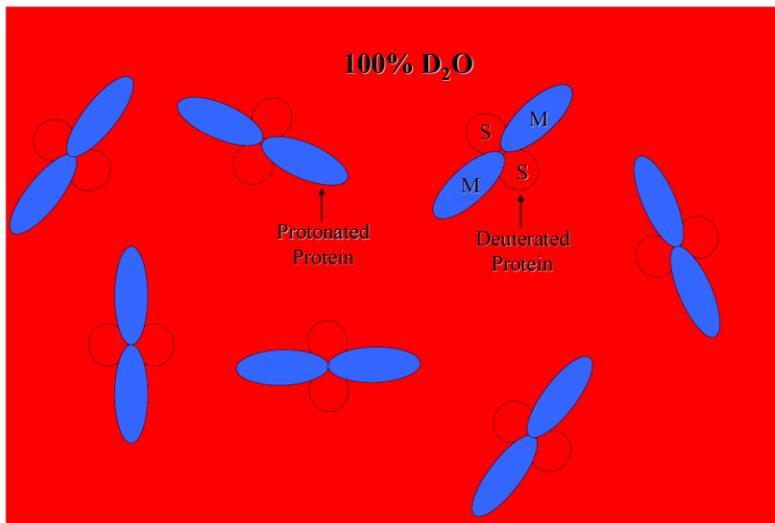
***h*-protein contrast matched at ~ 40% D₂O**

Contrast variation protein consisting of 2 sub-units

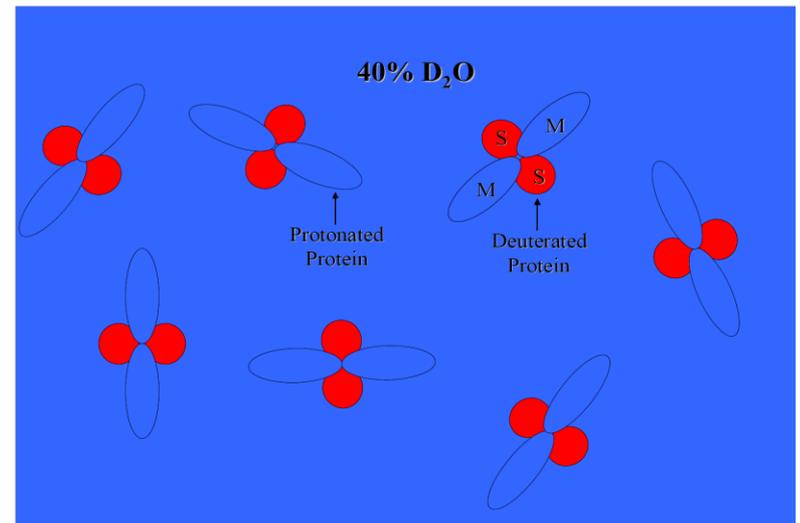


Hydrogenous
protein in D₂O

Protein with one sub-unit
deuterated in D₂O

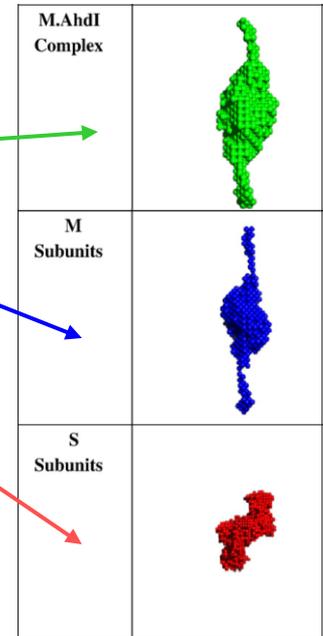
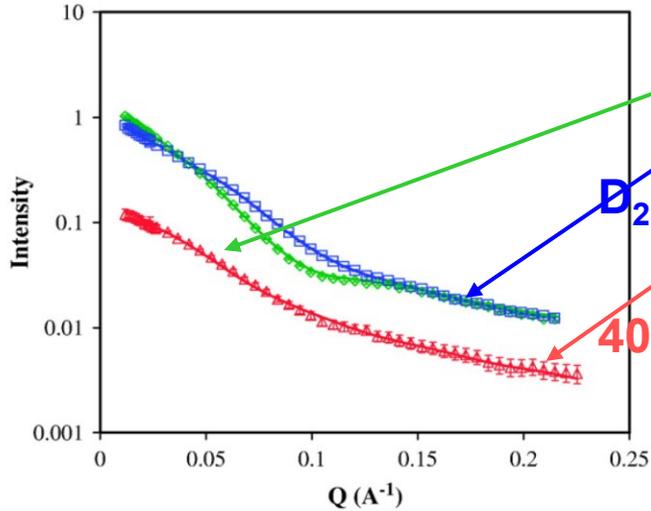


Protein with one sub-unit
deuterated in 40% D₂O

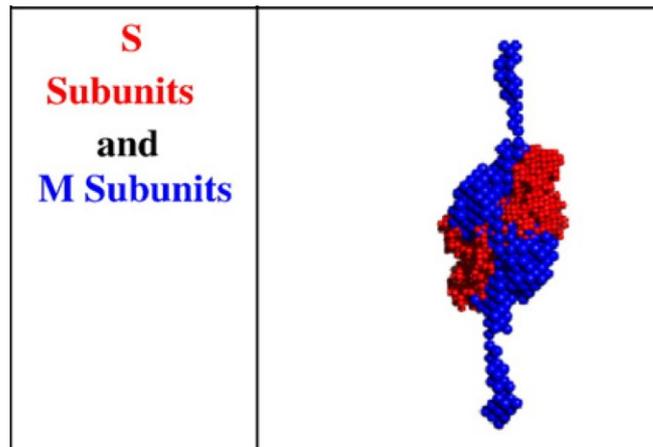


Shape and sub-unit organisation of the DNA methyltransferase M.AhdI by SANS

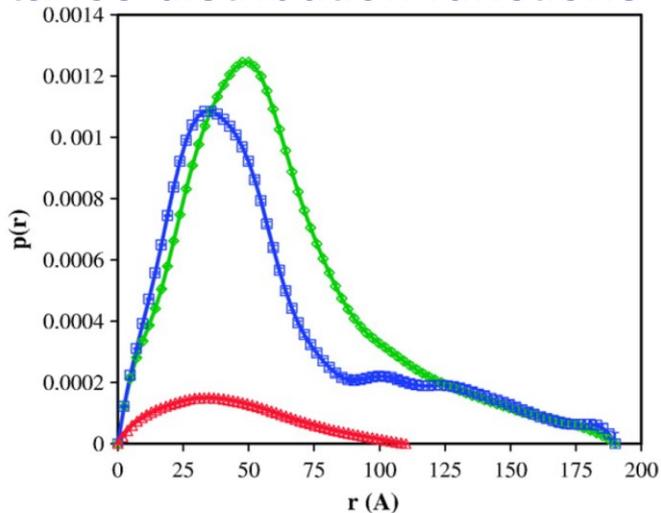
SANS data and *ab initio* fit (Svergun et al. 1999)



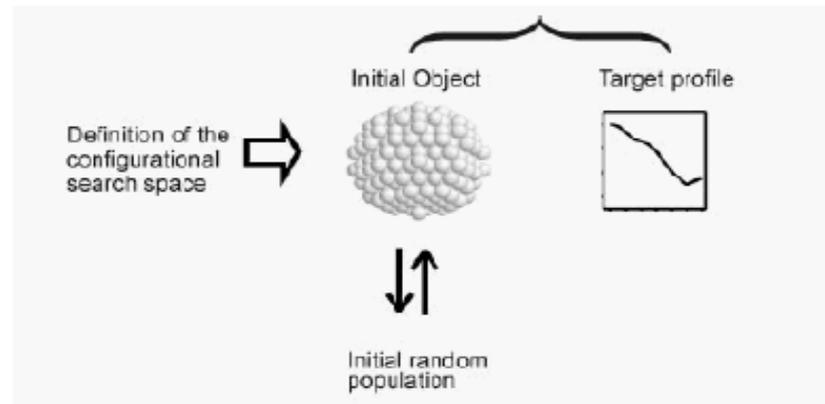
Model of M.AhdI



Distance distribution functions



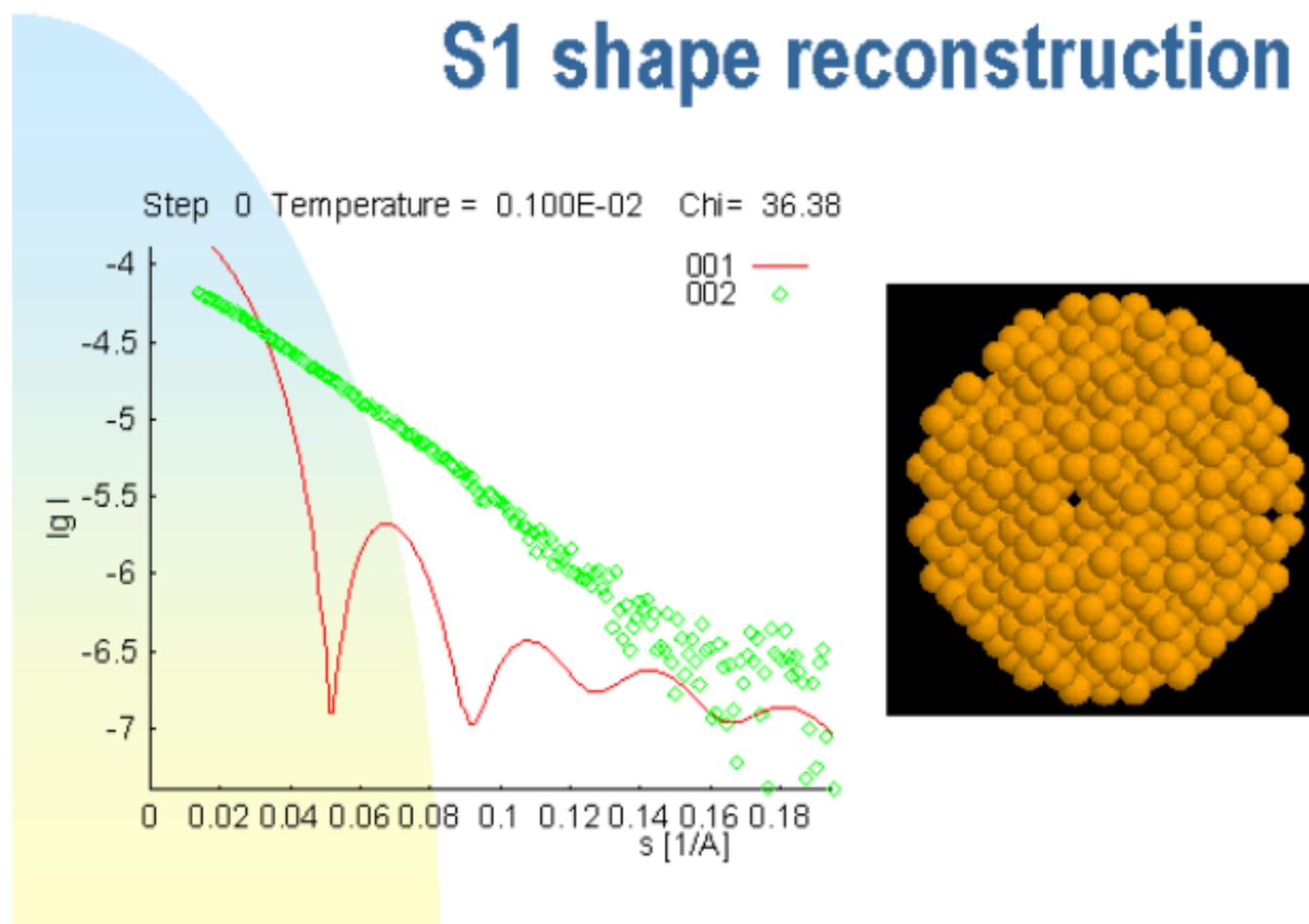
Ab initio modelling: an objective alternative to subjective approaches



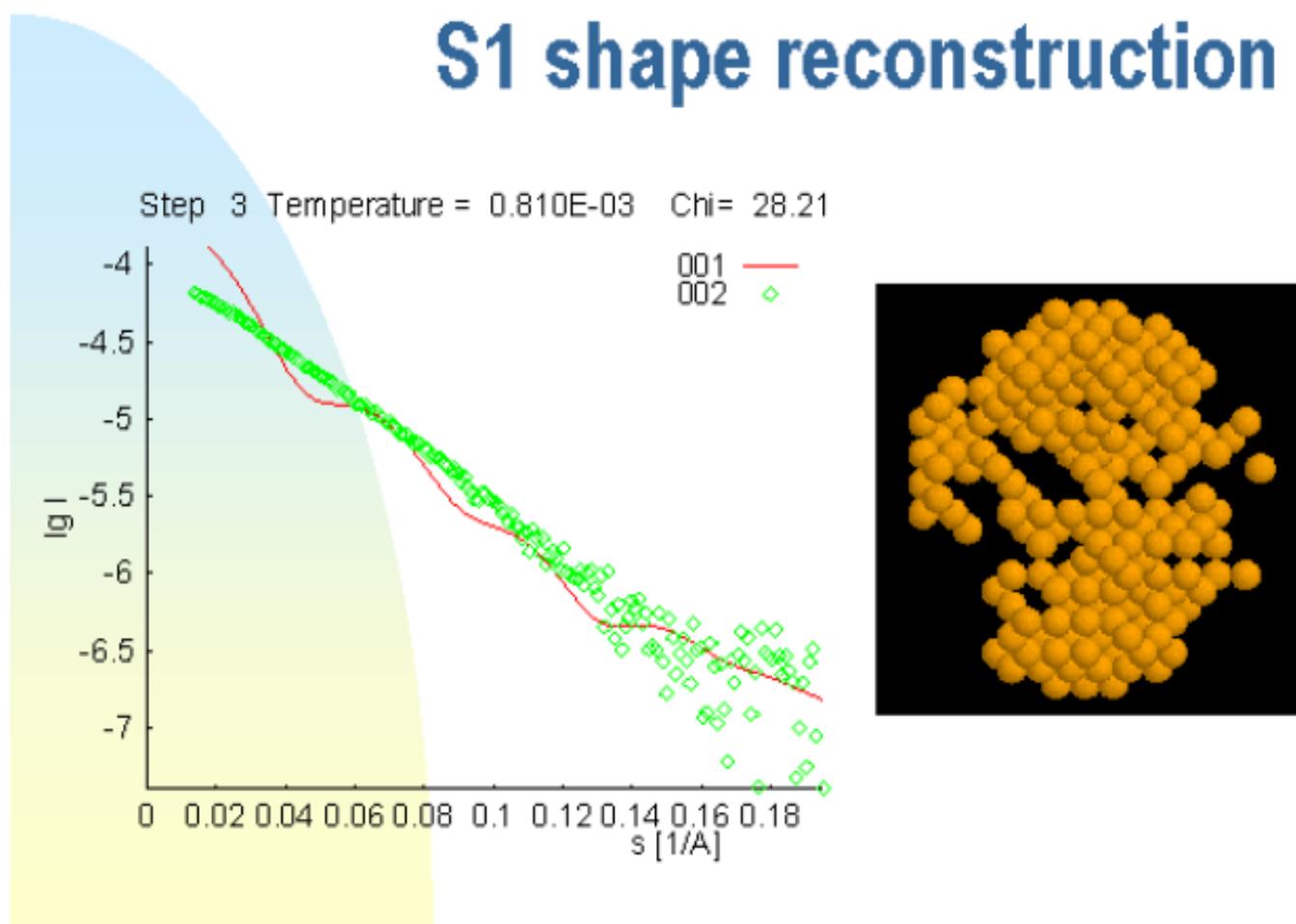
Ab initio shape determination by simulated annealing using a single phase dummy atom model

Dammin

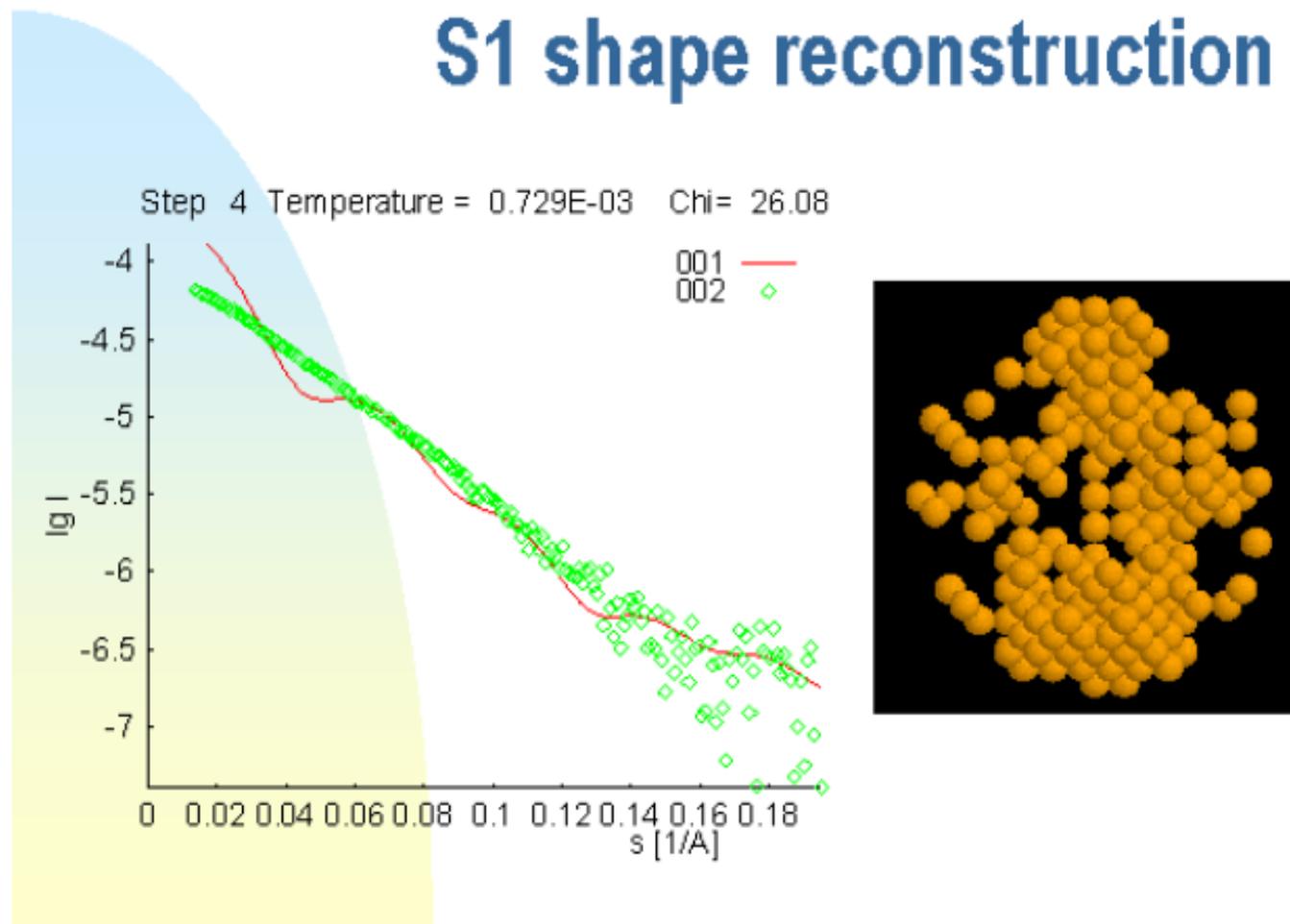
S1 shape reconstruction



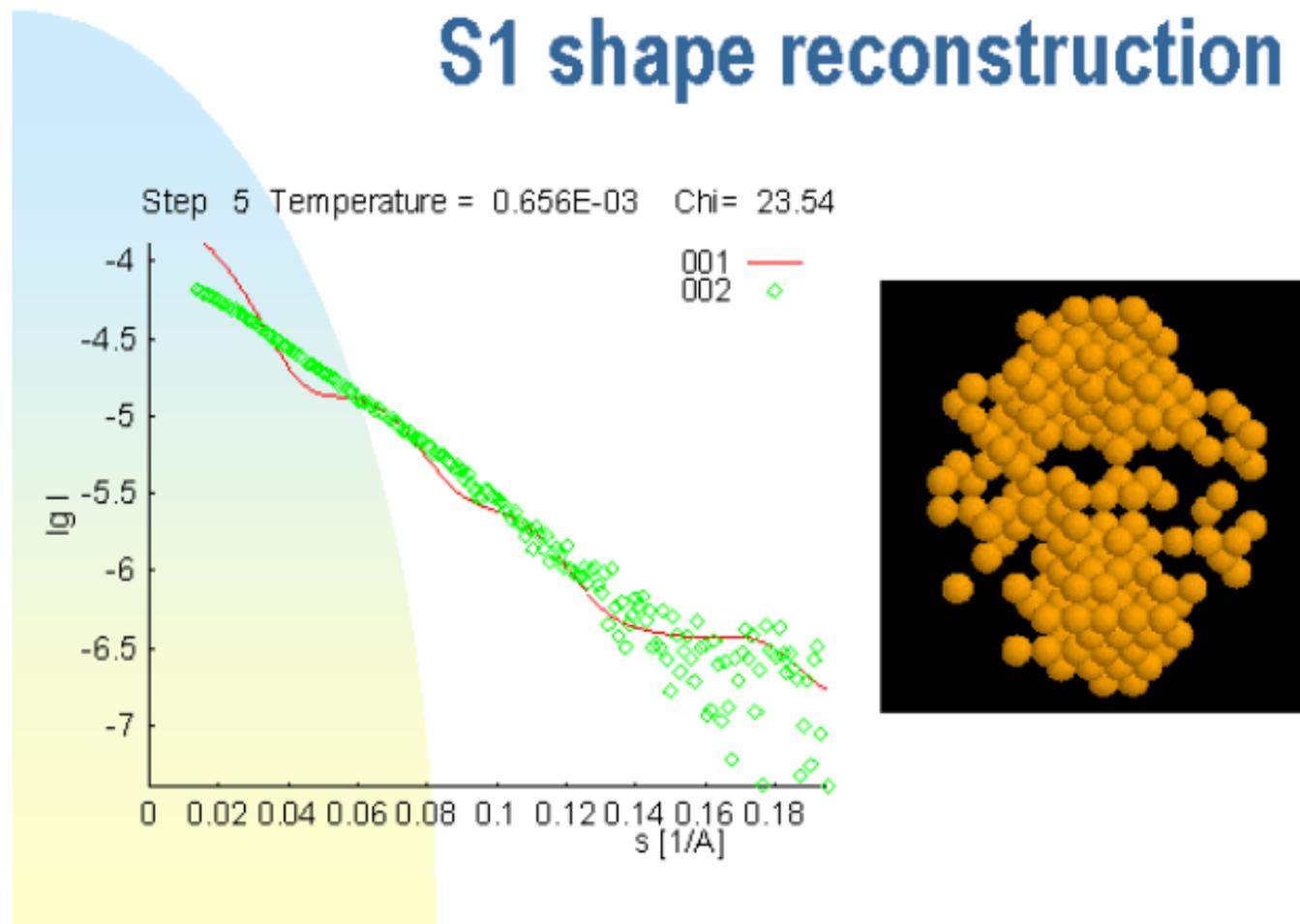
S1 shape reconstruction



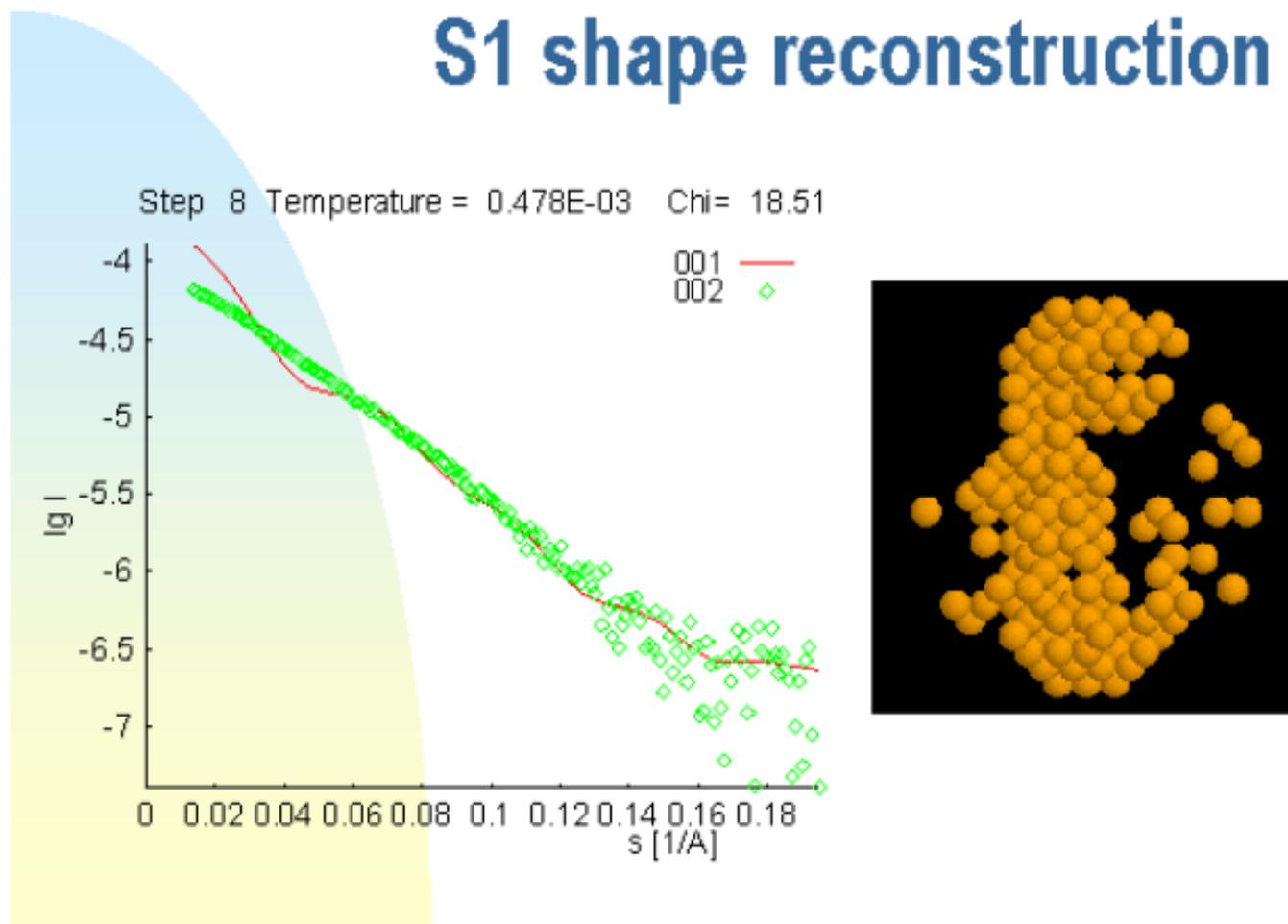
S1 shape reconstruction



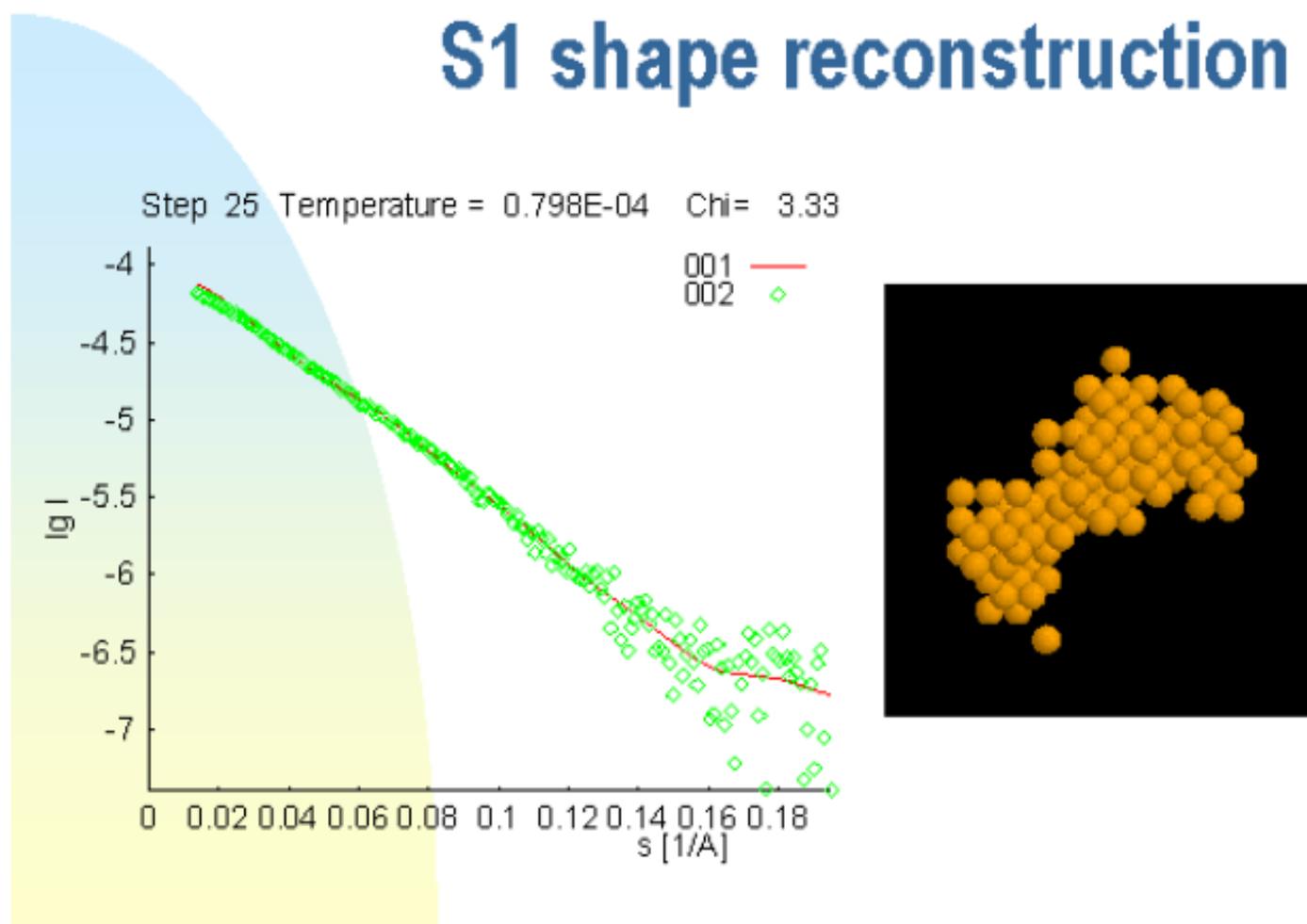
S1 shape reconstruction



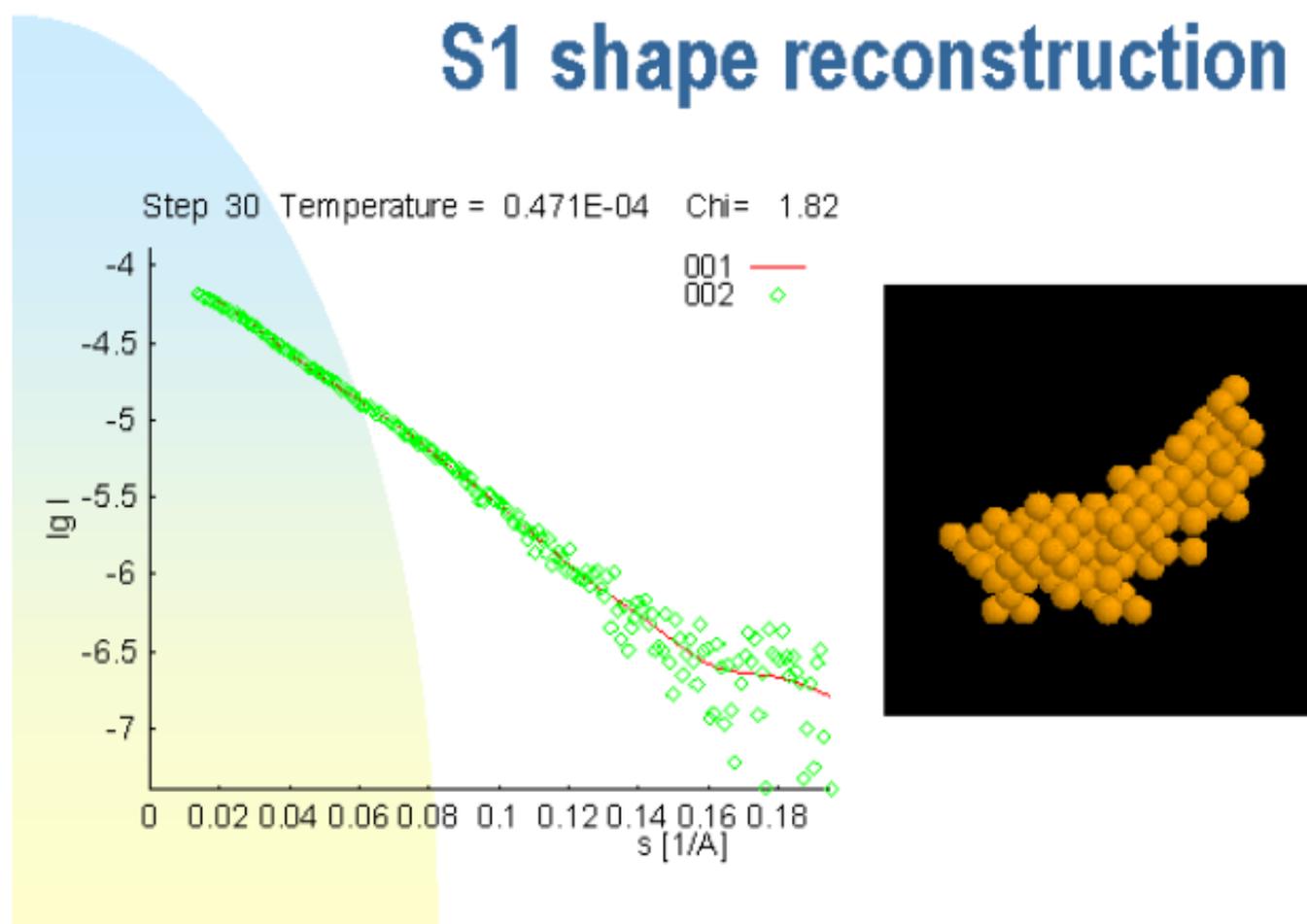
S1 shape reconstruction



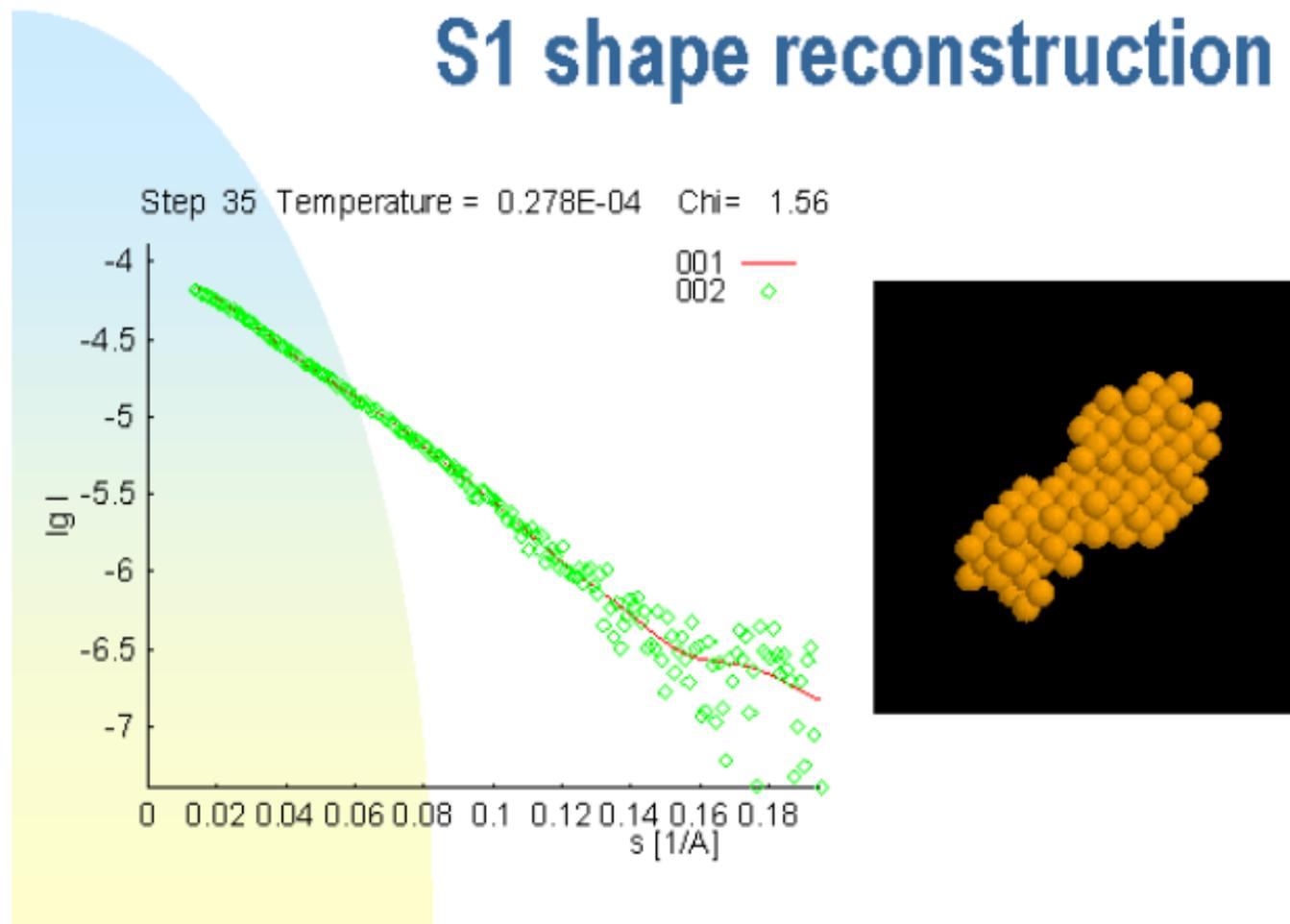
S1 shape reconstruction



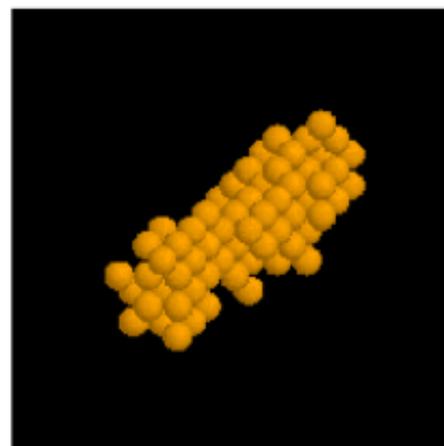
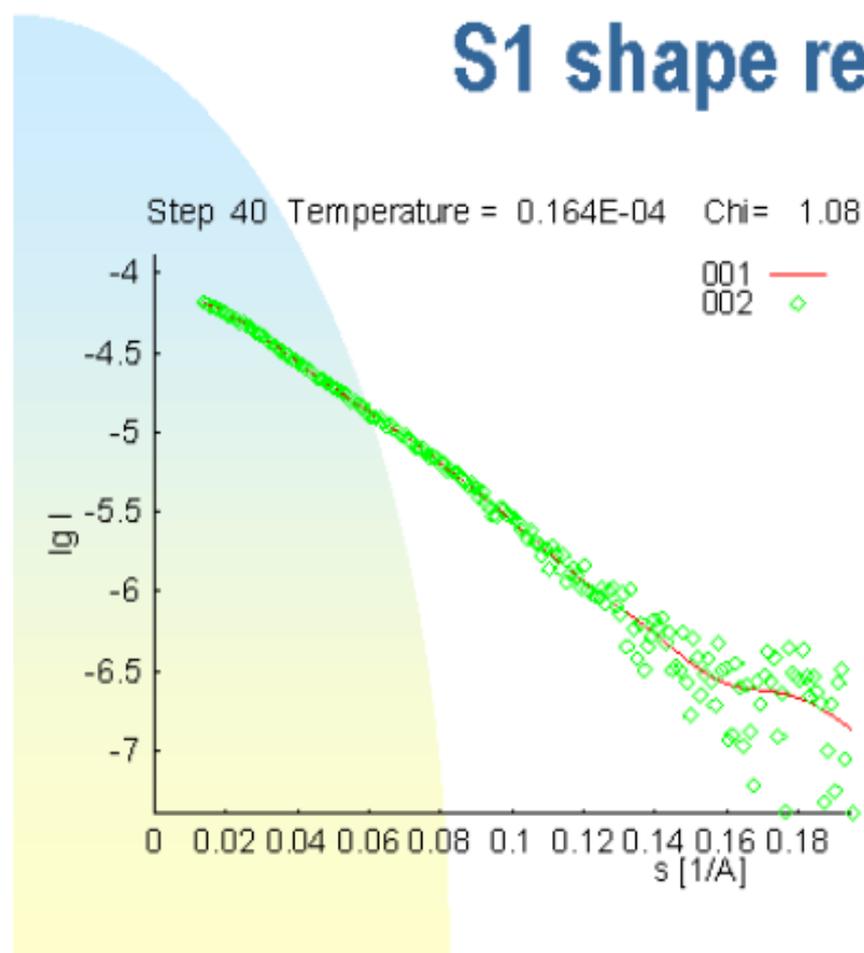
S1 shape reconstruction



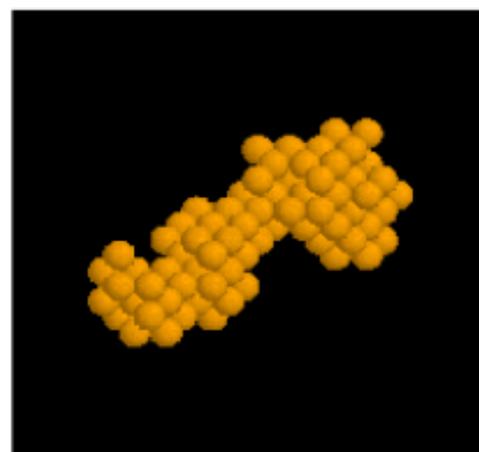
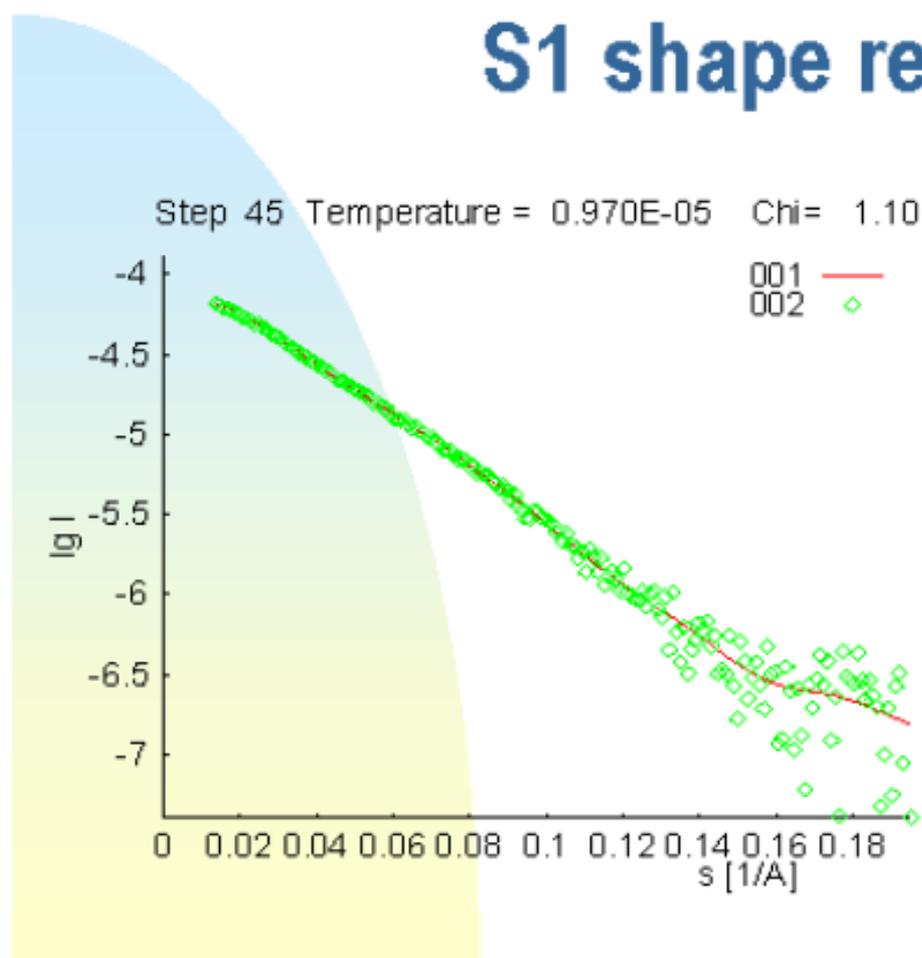
S1 shape reconstruction



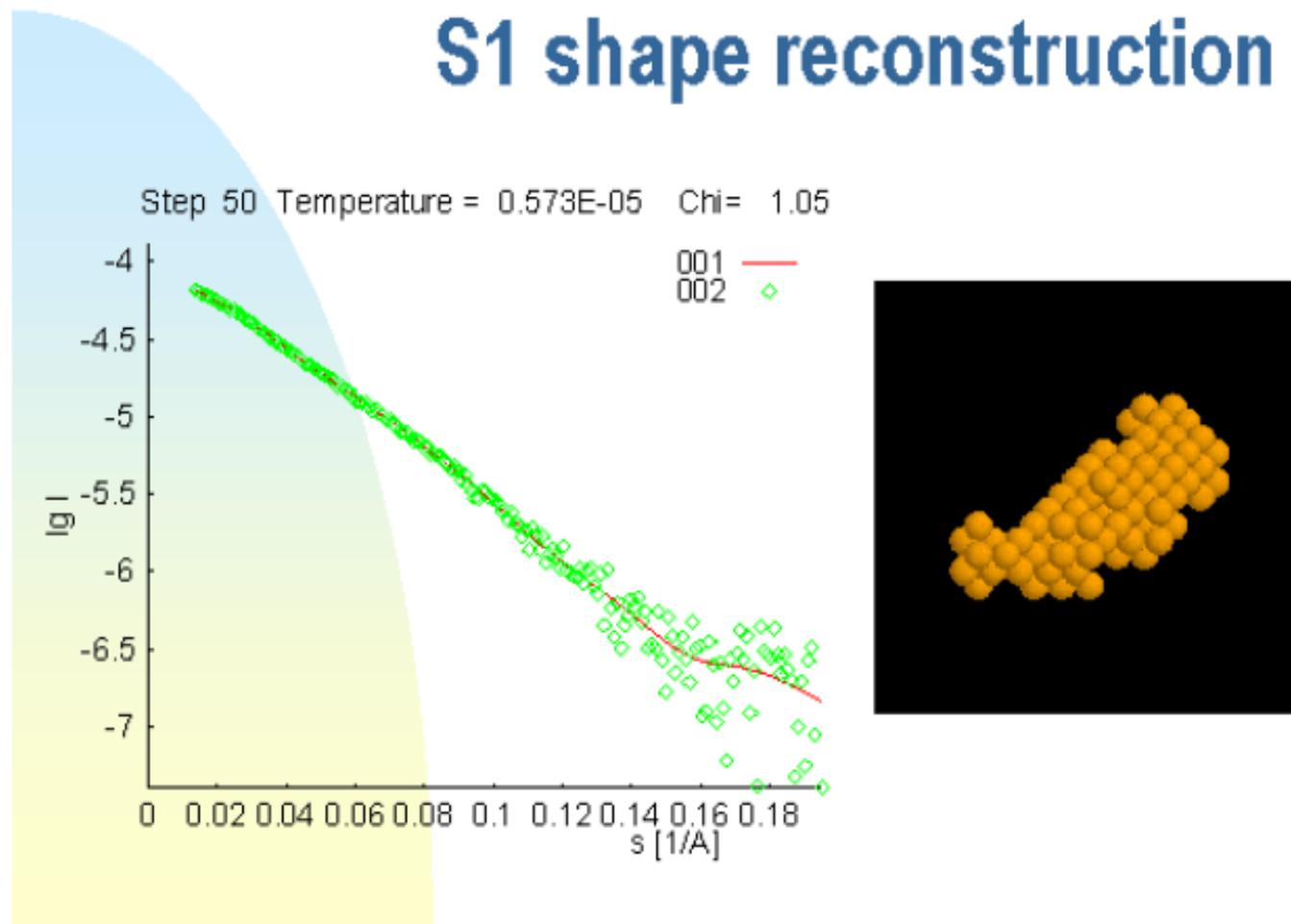
S1 shape reconstruction



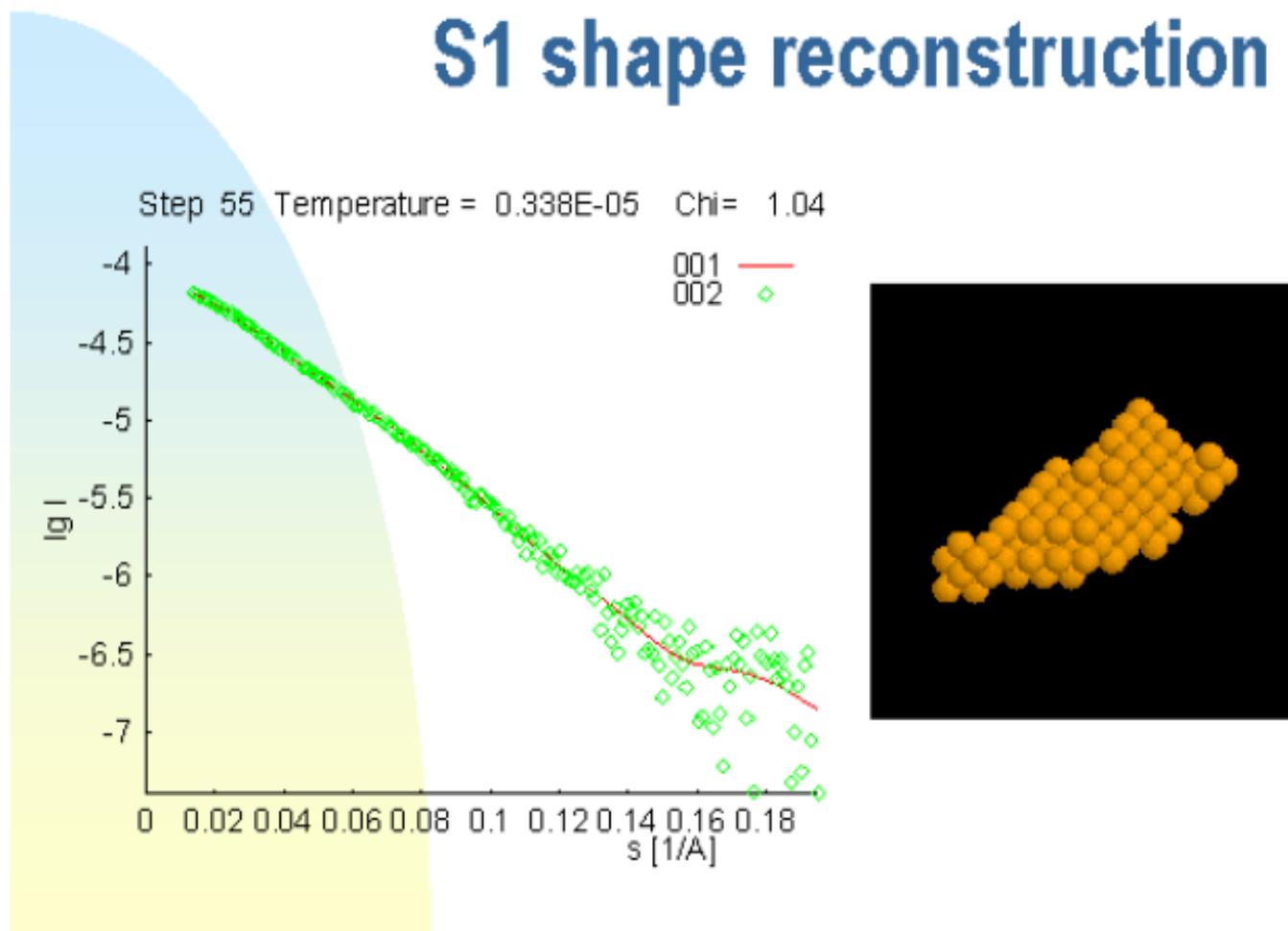
S1 shape reconstruction



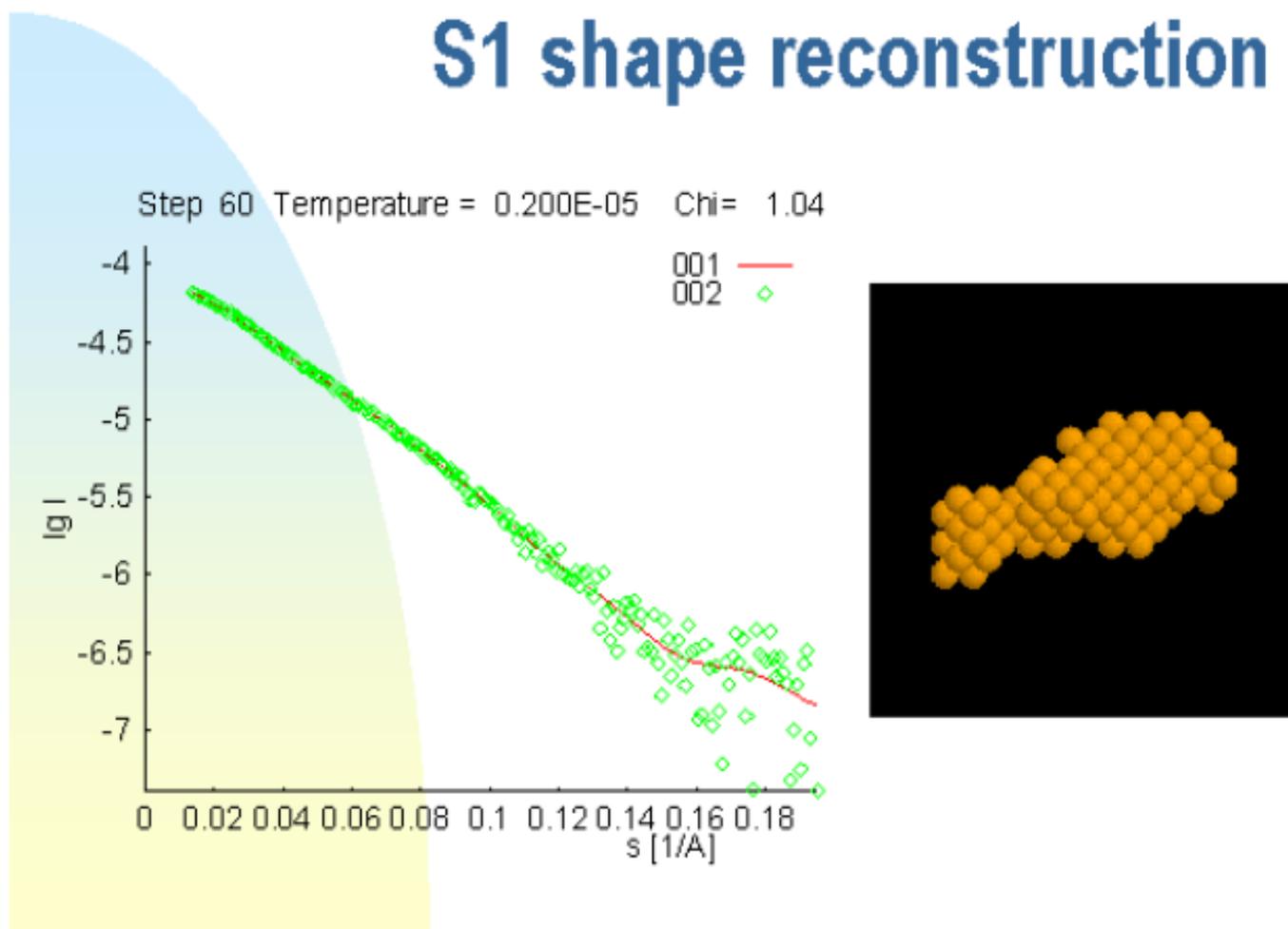
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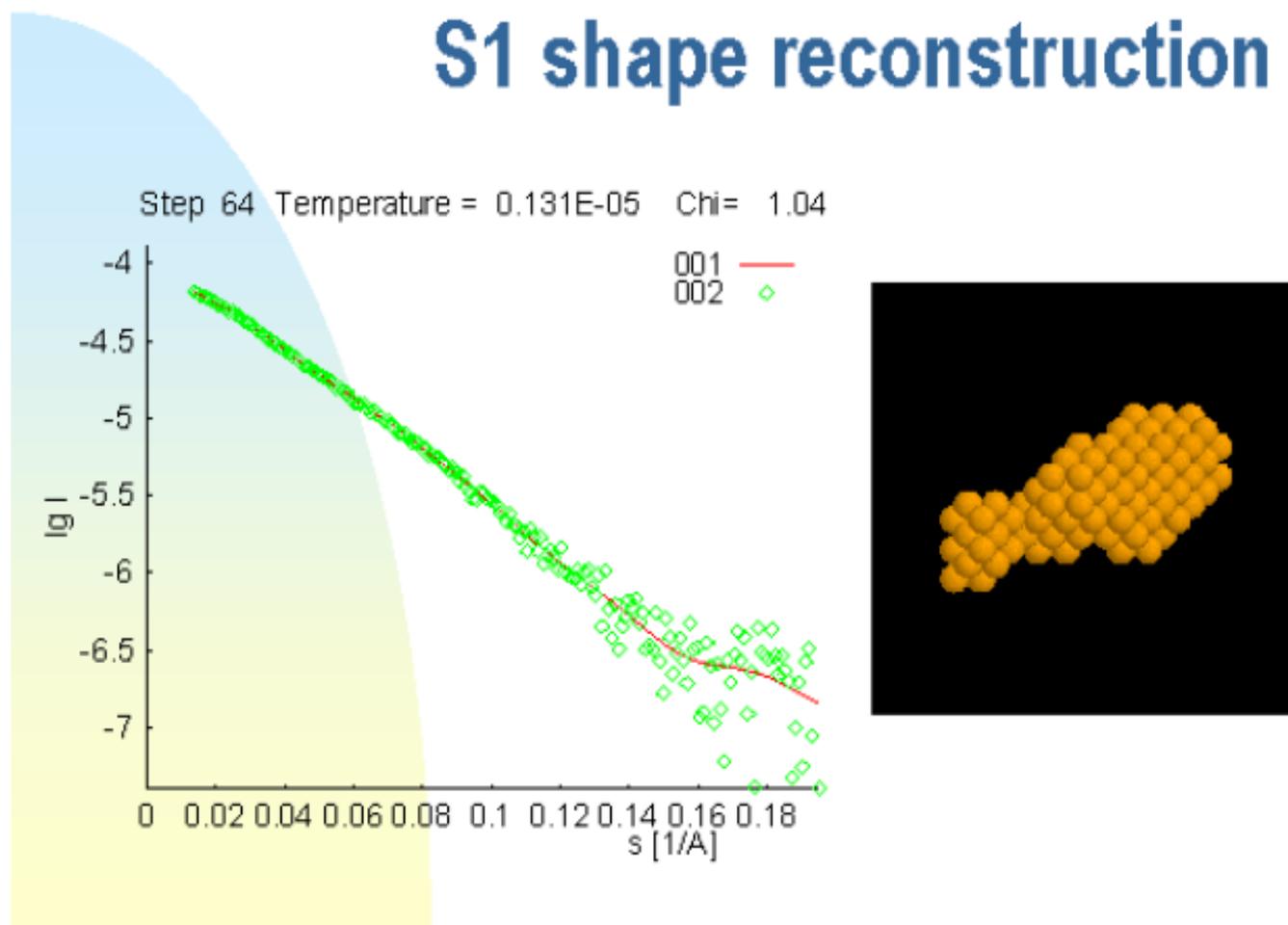
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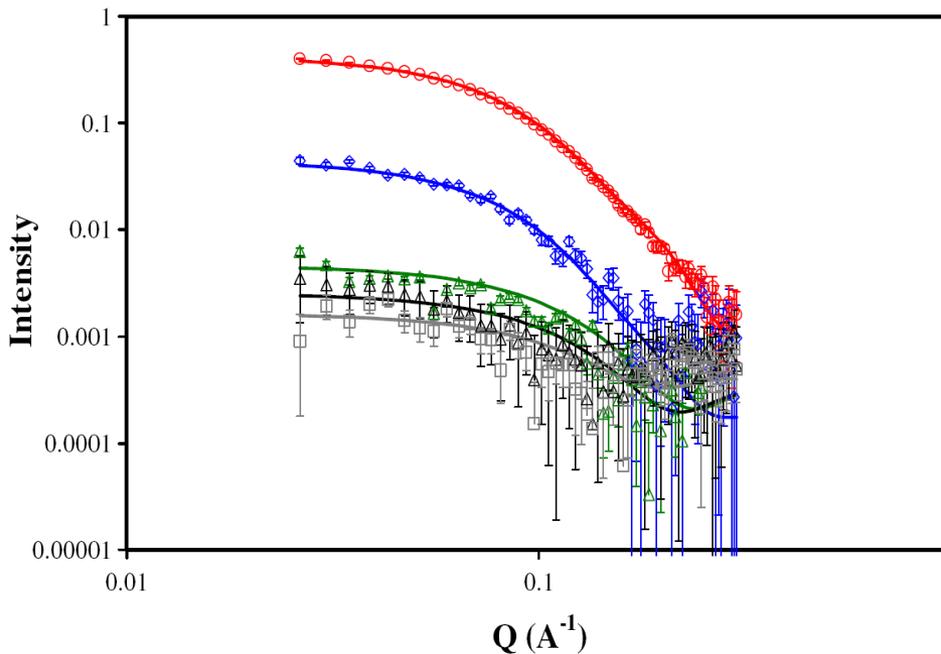
S1 shape reconstruction



S1 shape reconstruction



Selective deuteration of tryptophan & methionine in maltose binding protein (MBP)



Perdeuterated MBP in H₂O

Unmodified (h-)MBP in H₂O

Doubly labelled MBP trp. & met. residues deuterated (d-trp/met MBP) in 40% D₂O

Singly labelled MBP with trp. residues deuterated (d-trp MBP) in 40% D₂O,

Singly labelled MBP with met. residues deuterated (d-met MBP) in 40% D₂O

trp. and met. classified as very hydrophobic amino acid residues

8 trp. (d₃) residues and 8 met (d₅) out of a total of 370

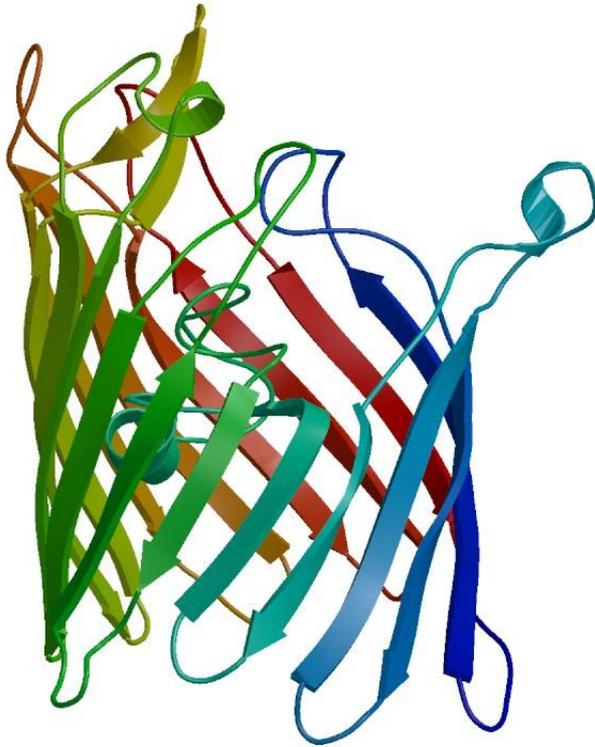
Low resolution neutron crystallography

- powerful technique for visualizing disordered regions in crystals of biomolecules (eg membrane proteins, viruses) and biomolecule complexes (eg DNA/protein complexes or lipoproteins)
- crystals can be less than 0.1 mm^3 in size
- if the sample is deuterium-labelled, crystals can be as small as 0.01 mm^3 due to better signal-to-noise ratios
- possible to highlight different components within a complex by complex matching using a crystallisation buffer of differing amounts of D_2O

Low resolution neutron crystallography

- provides important information about membrane proteins
- *membrane proteins* are usually removed from the cell membrane and crystallised using detergents which act to replace the lipid found *in vivo*
- information on the binding of these detergent molecules to the protein gives *indirect* information on the lipid binding in the membrane
- the detergent molecules are invisible in high resolution X-ray crystallographic electron density map due to their fluidity but can be visualized in low resolution neutron crystallography

Low resolution neutron crystallography



OmpF porin - integral membrane channel-forming protein which spans the outer membrane of Gram-negative bacteria (eg *E. coli*)

Allows the selective diffusion of small hydrophilic molecules across the membrane

Active pore trimer $\sim 75 \times 55 \text{Å}$

Low resolution neutron crystallography



View parallel to threefold axis of trimer

OmpF porin crystallised in 40% D₂O & therefore is contrast matched & 'invisible'

X-ray structure shown in yellow

deuterated detergent molecules are shown in purple

detergent molecules bound to 'hydrophobic zone' surrounding the trimer & which is exposed to lipid *in vivo*

Low resolution neutron crystallography

Results of investigations of membrane protein structures have demonstrated that:-

- not only are protein-protein interactions important in the formation of crystals
- but that in some cases protein-detergent and even detergent-detergent interactions are involved



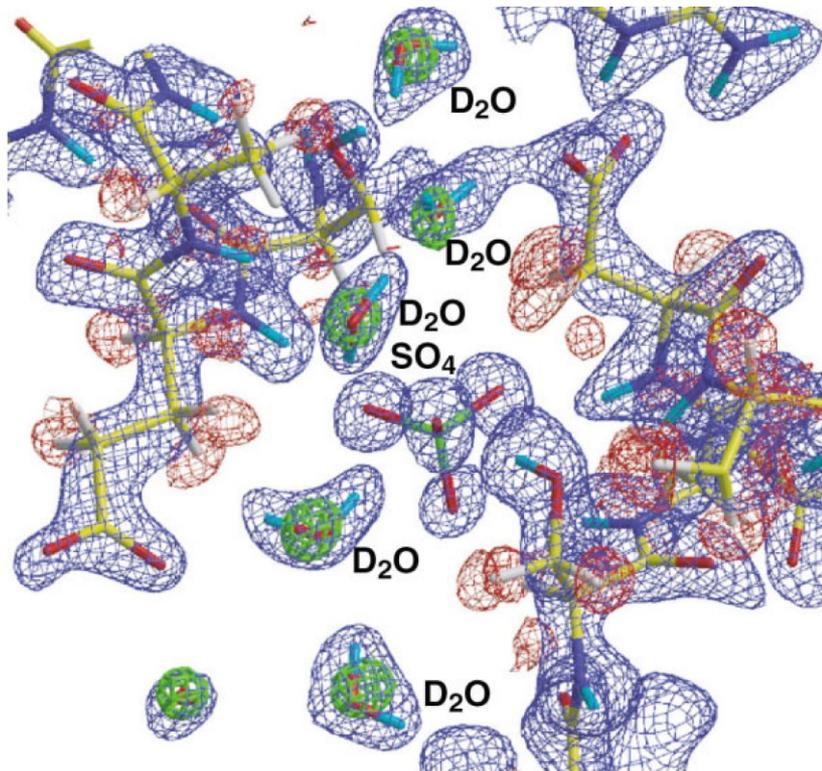
High resolution neutron crystallography - advantages

- Imaging of positions of water molecules bound to protein
 - Determination of hydrogen bonding networks
 - Both are critical to protein function as water mediates many biological functions
- Although X-ray crystallography and NMR spectroscopy are powerful methods for determining the 3D structure of proteins, it is difficult for such techniques to determine the positions of all the hydrogen atoms accurately

High resolution neutron crystallography - disadvantages

- Conventional high resolution neutron crystallography sees a low flux (until recently neutron crystallographic studies have been confined to monochromatic beam lines)
- High resolution protein crystallography was a minority interest due to the need for large crystals because of the low flux of neutron sources
- Very long collection times: weeks-months!
- Large crystals needed
- Problems with cryo-cooling

High resolution neutron crystallography - myoglobin



Myoglobin-myoglobin contact region

High-resolution neutron protein crystallography (@ 1.5Å resolution) provided information about

- the location of hydrogens including those present in water
- how the hydrogen in water is bound to neighbouring atoms and
- how the hydrogen in water is oriented (i.e., whether it packs in an ordered or disordered fashion)

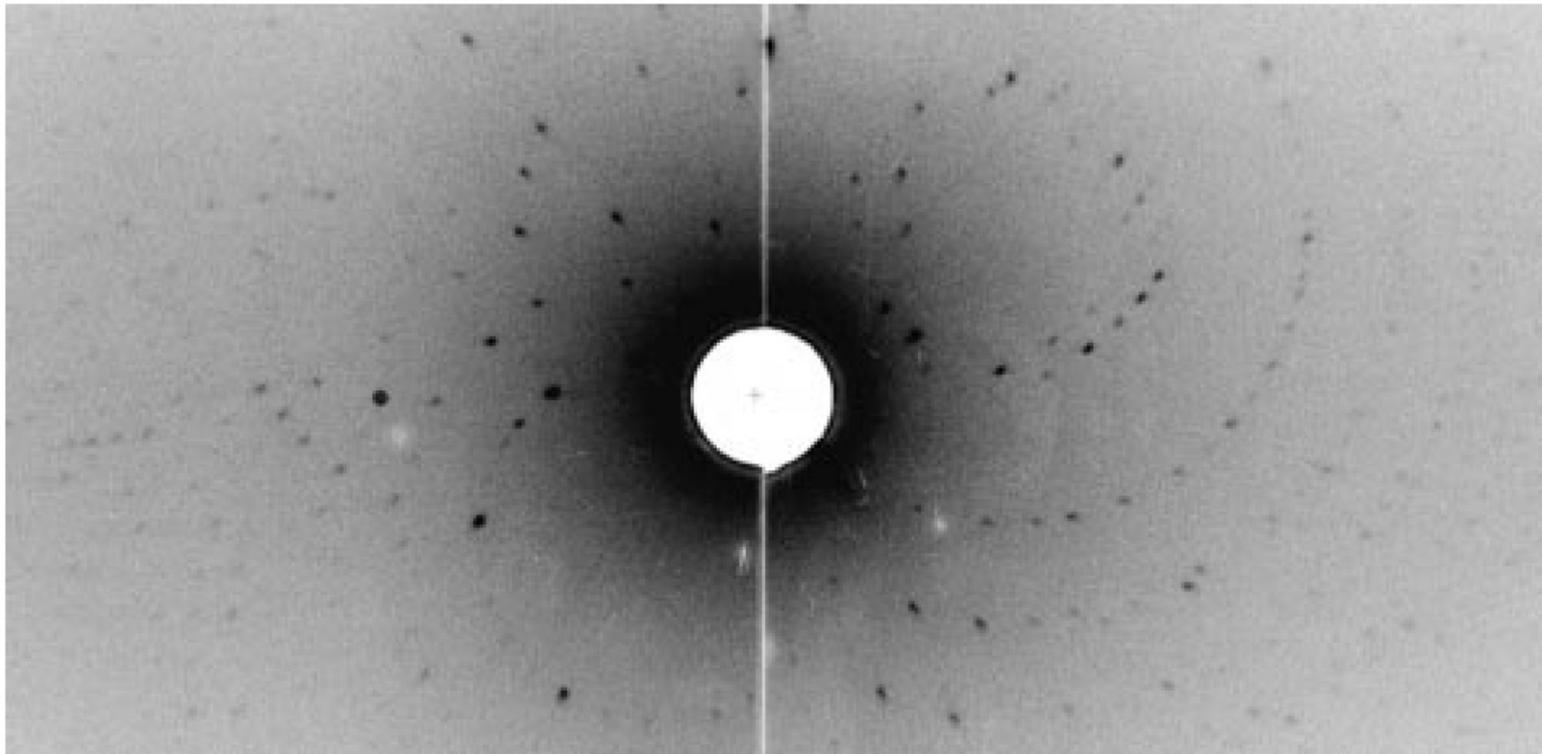
($2|F_o| - |F_c|$) neutron density map contoured at 1.5σ in blue and 2.0σ in red (note that hydrogen atom, ^1H , appears as a negative contour level due to its negative scattering length). The ($2|F_o| - |F_c|$) X-ray map for the water molecules is shown in green (the σ -levels for the water molecule density were individually set between 2.0 and 3.5 σ). The triangular neutron contours correspond to D_2O molecules.

LADI neutron Laue diffractometer at the ILL

- Illuminates sample with all available neutrons
- Maximal flux @ sample
- Large number of reflections at all incident neutron λ
- It is now routine for a protein crystal of 1-5 mm³ to be studied, although crystals as small as 0.15 mm³ have been recently measured on perdeuterated proteins
- Collection times reduced to hours-days. Data sets have been collected in 3.5 days to 2.0 Å resolution from a 1.4 mm³ thaumatin crystal and over several days to 2Å resolution from a perdeuterated antifreeze protein (AFP) only 0.13 mm³ in volume
- Upper limit of 50 kDa realisable

High resolution neutron crystallography – concanavalin A @ 15K*

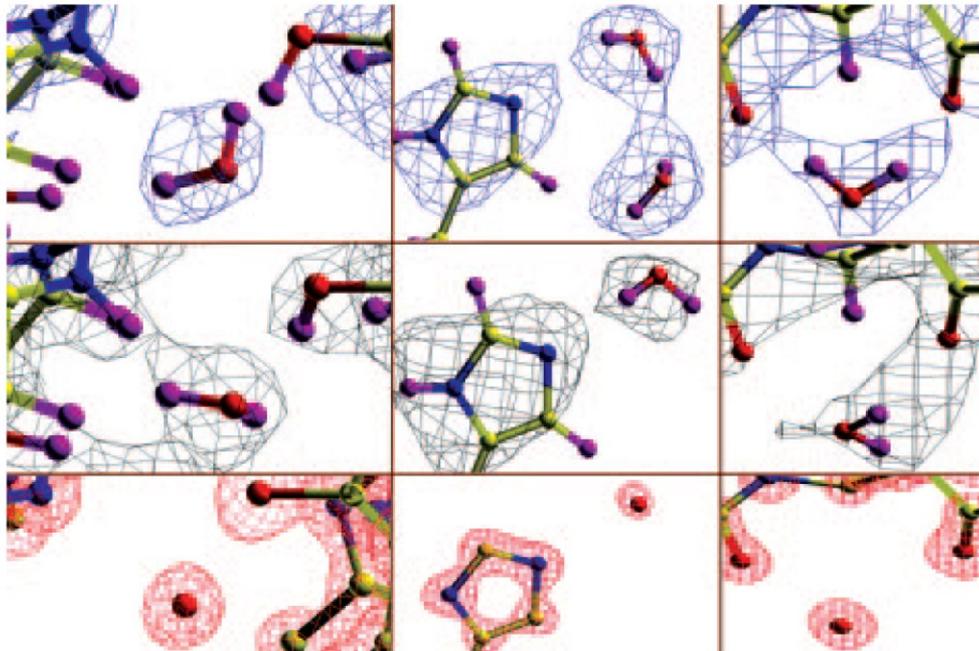
Concanavalin A is a saccharide-binding protein which belongs to the legume lectin family



1.5 mm³
Xystals

* to increase resolution

High resolution neutron crystallography – concanavalin A @ 15K

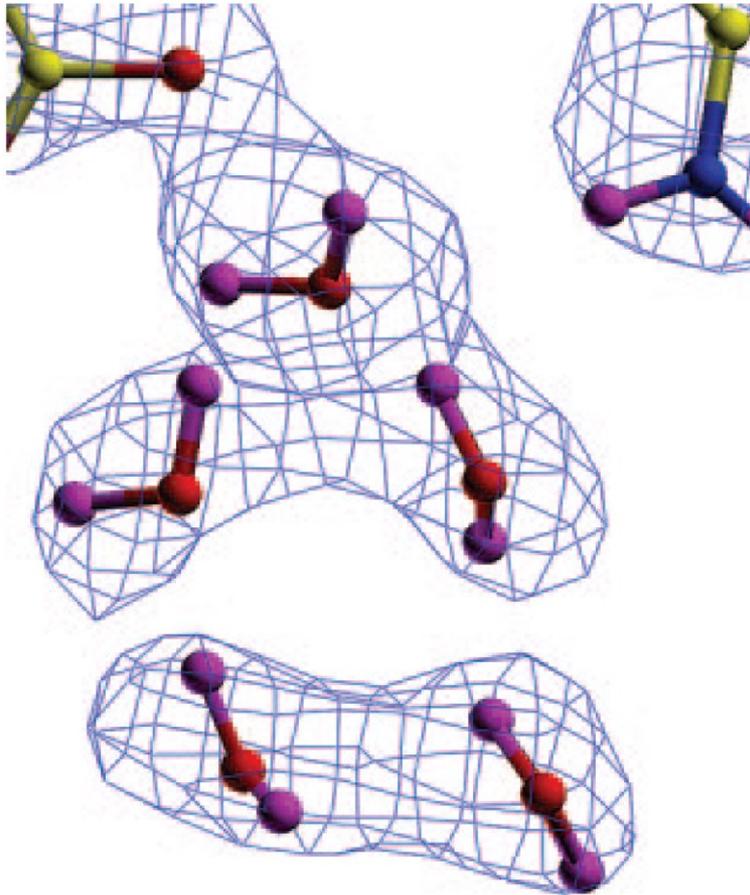


Neutron
crystallography
Neutron
crystallography at
293K

X-ray
crystallography at
111K

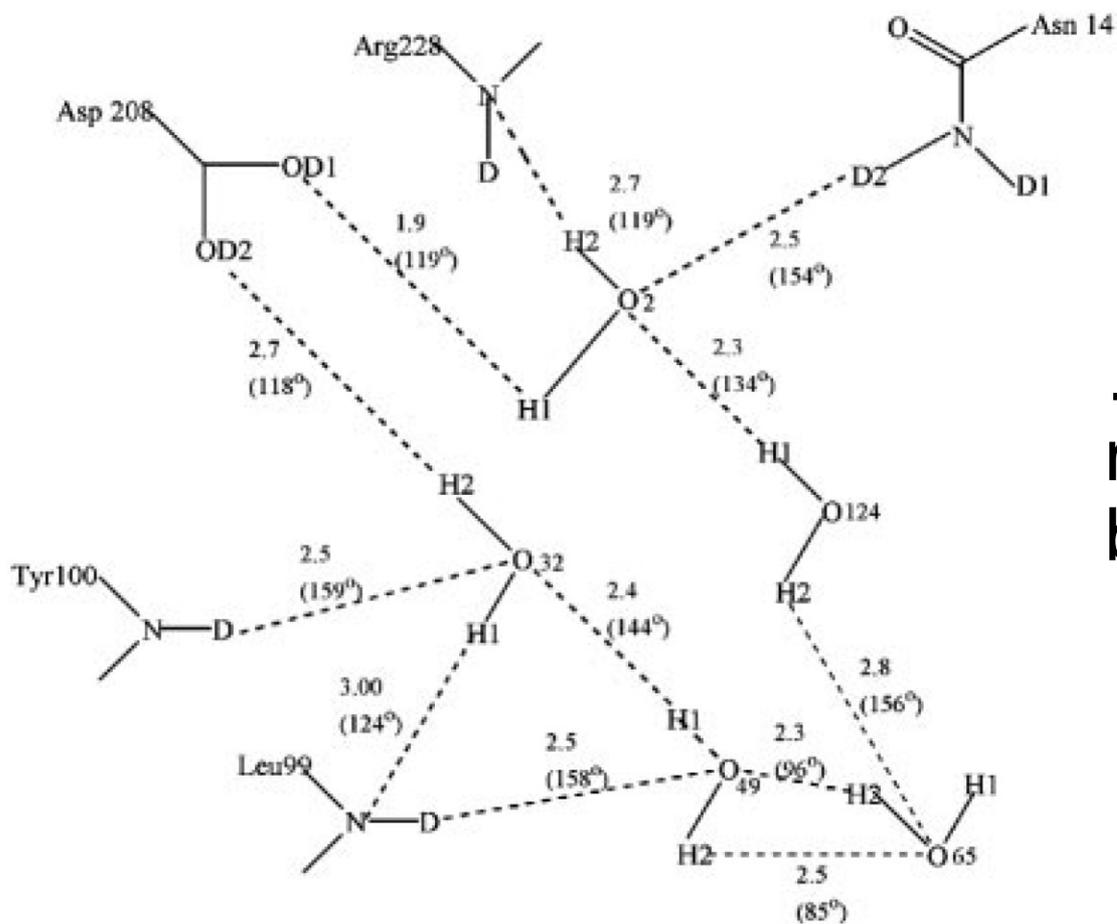
shows structure contains 2x bound water
molecules when compared to data
obtained at 293 K

High resolution neutron crystallography – concanavalin A @ 15K



and the positions of 5
waters in the saccharide-
binding site...

High resolution neutron crystallography – concanavalin A @ 15K



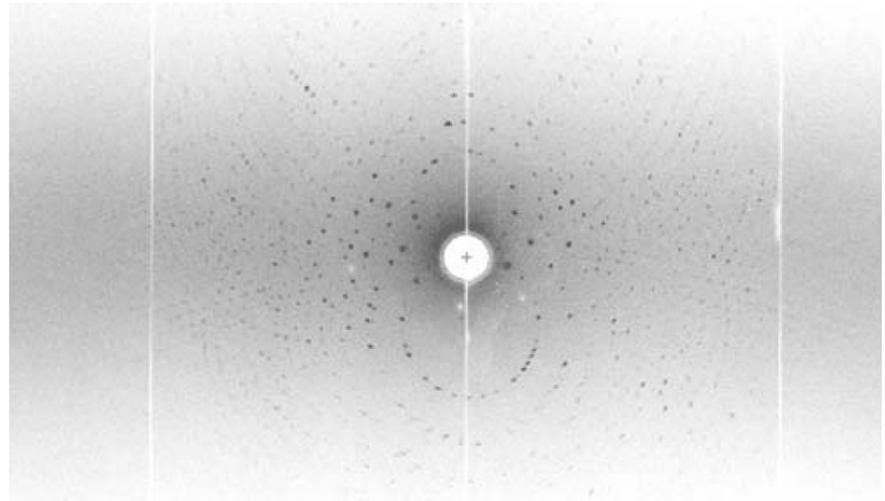
...and enables
resolution of the H-
bonding network

High resolution neutron crystallography

D-xylose isomerase:- 43 kDa enzyme that catalyses the first reaction in the catabolism of D-xylose

0.95Å X-ray structure but no direct observation of hydrogen although the refined model suggested that the site of ring opening was suitable for proton donation by His53.

Neutron laue diffraction data @ 2.2Å enabled direct observation of His53 protonation supporting the above mechanism



Neutron powder diffraction crystallography

- allows the real-space structure of materials to be determined at the atomic and micro-structural level
- allows determination of: long-range structure in polycrystalline materials, short-range atomic structure in disordered or amorphous materials, structural distortions, and any strain and crystal size induced changes to the structure
- it is complimentary to X-rays due to the neutrons' penetrative ability, light element sensitivity, isotope dependent scattering, and its magnetic interaction
- the ability of a neutron to penetrate materials allows the use of sophisticated sample environments - low and high temperatures, in electric and magnetic fields, and under varying pressure

Neutron powder diffraction crystallography

- Many pharmaceuticals contain predominately light elements such as H, D, C, N and O - the neutron's sensitivity to such elements and the difference in scattering between isotopes means neutron powder diffraction plays an important role in determining the structural features of such compounds